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Re-analysis of the (100), (001), and (020) rotational structure of SO_2 on the basis of high resolution FTIR spectra



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ABSTRACT

Three infrared spectra, weak (W), medium (M), and strong (S), of the ${}^{32}SO_2$ molecule were recorded with high resolution in the 1000–1500 cm⁻¹ region. Spectra were recorded with the Fourier Transform interferometer Bruker IFS-120 HR in Oulu (Finland) with different pressures, absorption path lengths, and recording time. That allowed us to record not only the ν_1 and ν_3 bands with higher values of quantum numbers J and K_a than it was made earlier, but to record for the first time very weak $2\nu_2$ band. In this case, transitions with the values $J^{max.}/K_a^{max.}$ equal to 89/37, 109/28, and 54/9 were assigned in the experimental spectra for the bands ν_1 , ν_3 , and $2\nu_2$, respectively. As it became clear in the course of the analysis, the rotational parameters of the ground vibrational state, known in the literature, do not describe suitably the ground state combination differences (GSCD) for the states with the value $K_a > 26-27$. As a consequence, the ground state rotational parameters were improved on the basis of our experimental data. The 12 131 transitions assigned in the experimental spectrum (7618, 3952, and 561 transitions of the bands ν_1 , ν_3 , and $2\nu_2$, respectively) were used for determination of ro-vibrational energy values of the vibrational states (100), (001), and (020). The lasts were used then in the fit procedure together with known in the literature high accurate sub-millimeter wave data. Resonance interactions between all three vibrational states have been taken into account in the Hamiltonian used for the fit. As a result, the 51 varied parameters, obtained from the fit, reproduce 4063 ro-vibrational energies of the states (100), (001), and (020) (12 131 initial experimental transitions) with accuracies close to experimental uncertainties: the rms deviation is 6.1×10^{-5} cm⁻¹, 9.7×10^{-5} cm⁻¹, and 13.9×10^{-5} cm⁻¹ for our FTIR data (for the (100), (001), and (020) states, respectively), and comparable with experimental uncertainties for heterodyne data.

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1. Introduction

Sulfur dioxide is an important chemical species in many fields such as chemistry, astrophysics, laser techniques. It

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plays an important role as a pollutant in the terrestrial atmosphere. In particular, sulfur dioxide is one of the major air pollutants released in the atmosphere as a result of volcanic eruptions and of fuel combustion in human activities; it contributes to the generation of smog and constitutes a serious health hazard for the respiratory system. To solve the problems of propagation of monochromatic radiation in the atmosphere, laser sounding, information transfer,

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the remote detection and monitoring of SO_2 in situ, etc., one also should have a good knowledge of the fine structure of the SO_2 absorption spectra in different parts of the electromagnetic spectrum, in particular, in the infrared. Therefore, spectroscopic studies of the sulfur dioxide molecule have been made during many years both in the microwave, and sub-millimeter wave and infrared regions (see, Refs. [1–35].

In this paper we follow up on our recent studies of the high resolution spectra of SO₂, Refs. [36-40]. The subject of the present study is the region of $1000-1500 \text{ cm}^{-1}$. This region, where the strongest SO₂ ν_3 and ν_1 bands, and very weak $2\nu_2$ band are located, is ideal for infrared atmospheric measurements of that molecule. Earlier the ν_1 and ν_3 bands were discussed in Refs. [5,12,16,21,23]. All information containing in these references were then generalized in Ref. [26]. But the information about the band $2\nu_2$ was absent, and ro-vibrational energies of the (020) state were determined from the hot $2\nu_2 - \nu_2$ band [21]. In our present analysis, we were able to assign transitions of the ν_1 and ν_3 bands with values of quantum numbers J and K_a considerably higher than in Ref. [26]. Moreover, transitions of the very weak $2\nu_2$ band also were recorded and assigned for the first time. Section 2 describes the experimental conditions for the recording spectra. In Section 3 we briefly present the Hamiltonian model used to fit the experimental line positions. Description of the spectra, assignment of transitions, and the problem of improvement of the ground state rotational parameters are discussed in Section 4. The results of analysis of the high resolution spectra of the bands ν_1 , ν_3 and $2\nu_2$ and determination of spectroscopic

parameters of the states (100), (001), and (020) are presented in Section 5.

2. Experimental details

For this molecular research three infrared spectra (weak, medium, and strong) of the ${}^{32}SO_2$ molecule in the region from 1000 to 1500 cm⁻¹ were recorded with a high resolution Fourier Transform spectrometer Bruker IFS-120 HR in the infrared laboratory at the University of Oulu (Finland). In these measurements two absorption cells were used. In the W and M registrations the SO₂ sample, made by Sigma-Aldrich Inc., was introduced in a White type cell, Ref. [41], at the pressure of 11.7 and 110.7 Pa, respectively. The absorption path length (APL) was 3.2 m. In the S measurement the sample was at pressure of 110.7 Pa in another cell, Ref. [42], with APL=163.2 m. The both cells have been optimized for the used spectrometer, and they were provided with potassium bromide (KBr) windows in these measurements. The instrumental resolution at 1100 cm⁻¹ was better than 0.0019 cm⁻¹ in all three registrations. A globar source, a germanium film between two KBr plates, as the beamsplitter, and mercury cadmium telluride (MCT) detector were used. The registration times were 42.4, 79.9 and 99.9 h in the W, M and S measurements, respectively. The spectra were calibrated with peaks of the OCS $2\nu_2$ band [43] measured in fourth registration together with the sample spectrum. The peak positions were calculated with an optimized center of gravity method [44].



Fig. 1. Survey weak (W) spectrum of SO₂ in the region of 1050–1450 cm⁻¹. Experimentally recorded strong bands ν_1 and ν_3 can be seen in the upper part of the figure. Experimental conditions: absorption path length is 3.2 m; sample pressure is 11.7 Pa; room temperature; registration times is 42.4 h. Synthetic spectrum is shown on the lower part of the figure. In the 1400–1450 cm⁻¹ spectral range, some lines that are appeared in the experimental spectrum but not in the calculated one, belong to H₂O.



Fig. 2. Survey strong (S) spectrum of SO₂ in the region of the ν_1 and $2\nu_2$ bands. The P-branch of the weak $2\nu_2$ band is seen in the left part of the figure. Experimental conditions: absorption path length is 163.2 m; sample pressure is 110.7 Pa; room temperature; registration times is 99.9 h.



Fig. 3. A small part of the high resolution spectrum of SO₂ in the region of *R*-branch of the ν_1 band (upper part). Experimental conditions correspond to the weak spectrum W. Lines of ³²SO₂ are marked by dark circle. Some weak lines which are absent in the synthetic spectrum (lower part), belong to the hot band $\nu_1 + \nu_2$ (they are marked by dark square), or to the ³⁴SO₂ isotopic species (they are marked by dark triangle).

3. Hamiltonian model

The SO₂ molecule is an asymmetric top of the $C_{2\nu}$ symmetry. On this reason its three vibrational modes possess the following symmetry: $q_{\lambda} \in A_1$ for $\lambda = 1, 2$, and

 $q_{\lambda} \in B_1$ for $\lambda = 3$. As a consequence, two type bands are allowed in absorption: parallel (or A_1) type bands that correspond to the vibrational transitions $(v_{A_1}) \leftarrow (v_{gr})$ and perpendicular (or B_1) type bands that correspond to the vibrational transitions $(v_{B_1}) \leftarrow (v_{gr})$. The selection rules for

Table 1High accurate sub-millimeter wave transitions of the SO_2 molecule.

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	0.3 -1.0 -16.2 -7.0 8.0 0.9 1.5 10.0 16.1 -0.5	$2.2 \\ 4.4 \\ -5.1 \\ 2.5 \\ 5.3 \\ -0.6 \\ 1.9 \\ 4.6 \\ 4.2$
70 63 62 16 7 63 $42505.427(10)$ 68 8 60 \leftarrow 68 7 61 $496010.225(15)$ 69 8 62 \leftarrow 68 9 59 $498109.421(20)$ 49 15 35 \leftarrow 50 14 36 $498657.914(10)$ 70 9 61 \leftarrow 69 10 60 $498920.109(20)$ 72 8 64 \leftarrow 72 7 65 $499505.309(10)$ 54 16 38 \leftarrow 55 15 41 $500111.014(10)$ 59 17 43 \leftarrow 60 16 44 $501179.560(15)$ 64 18 46 \leftarrow 65 17 49 $501871.632(20)$	-1.0 -16.2 -7.0 8.0 0.9 1.5 10.0 16.1 -0.5	4.4 -5.1 2.5 5.3 -0.6 1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	-162 -7.0 8.0 0.9 1.5 10.0 16.1 -0.5	-5.1 2.5 5.3 -0.6 1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	-7.0 8.0 0.9 1.5 10.0 16.1 -0.5	2.5 5.3 -0.6 1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	8.0 0.9 1.5 10.0 16.1 -0.5	5.3 -0.6 1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	0.9 1.5 10.0 16.1 0.5	-0.6 1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1.5 10.0 16.1 -0.5	1.9 4.6 4.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10.0 16.1 0.5	4.6
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	16.1 -0.5	42
	-0.5	
53 16 $38 \leftarrow 54$ 15 39 $519365.544(10)$		2.0
58 17 41 \leftarrow 59 16 44 520 382 480(15)	3.8	16
63 18 46 \leftarrow 64 17 47 521011924(20)	15.4	8.6
73 20 54 \leftarrow 74 19 55 521 137.011(100)	42.9	-10.3
$68 19 49 \leftarrow 69 18 52 521261.295(40)$	15.4	-4.0
47 15 33 \leftarrow 48 14 34 537 227.58(10)	-11.9	1.2
52 16 36 \leftarrow 53 15 39 538 602.765(10)	-3.4	1.3
77 7 71 \leftarrow 78 4 74 539.061.368(60)	-13.6	21.8
57 17 41 \leftarrow 58 16 42 539 570.174(10)	2.4	3.0
78 6 $72 \leftarrow 79$ 5 75 $539807.786(100)$	-9.2	22.4
72 20 52 \leftarrow 73 19 55 540112.736(40)	34.9	-3.8
62 18 44 \leftarrow 63 17 47 540 139.119(20)	9.6	7.0
75 9 67 \leftarrow 74 10 64 543 973.462(10)	-11.9	-6.3
$80 9 71 \leftarrow 80 8 72 548 \ 752 \ 732 \ (20)$	-15.6	-8.7
76 8 68 \leftarrow 76 7 69 549.848.964(10)	-1.4	-9.7
78 9 69 \leftarrow 78 8 70 551403.734(40)	13.3	22.8
41 7 35 \leftarrow 42 4 38 564 029.767(15)	-28.2	2.5
51 8 44 \leftarrow 52 5 47 564 469.319(10)	-20.6	6.4
72 9 63 \leftarrow 71 10 62 564 665.735(10)	10.0	2.0
76 9 67 \leftarrow 76 8 68 564 896,451(20)	-9.0	2.2
60 4 56 \leftarrow 61 3 59 567 304.250(20)	8.9	-1.8
70 7 63 \leftarrow 70 6 64 569 387.048(10)	6.3	-6.9
$69 \qquad 6 \qquad 64 \qquad \leftarrow \qquad 70 \qquad 3 \qquad 67 \qquad 571153.962(20)$	-7.3	-6.8
51 4 48 \leftarrow 52 1 51 574 052.463(10)	6.1	1.2
50 16 34 \leftarrow 51 15 37 577 028.807(10)	-9.6	-1.4
79 7 73 \leftarrow 80 4 76 577 256,699(60)	14.7	56.4
55 17 39 \leftarrow 56 16 40 577 902.912(10)	-2.5	2.6
70 20 50 \leftarrow 71 19 53 578 039.819(30)	1.9	-13.8
60 18 42 \leftarrow 61 17 45 578 356.855(15)	-4.6	-0.1
65 19 47 \leftarrow 66 18 48 $578399.767(20)$	-22.9	-22.1
70 5 65 \leftarrow 71 4 68 579 237.260(20)	-1.9	-0.8
84 9 75 ← 84 8 76 579 739.58 (30)	8.6	9.1

Table 1 (continued)

	ominueu)								
Transitio	on						Frequency, exp. ^a in MHz	Δ (calc) ^b in kHz	∆ (calc) ^c in kHz
1							2	3	4
40	2						500 (10 000/15)		
43	3	41	<i>←</i>	44	0	44	580 618.830(15)	-2.7	0.5
66	8	58	←	65	9	57	582 021.341(10)	7.4	-3.1
74	9	65	←	74	8	66	587 316.386(10)	-9.4	3.3
52	3	49	←	53	2	52	587 329.413(20)	-2.1	-6.6
60	8	52	\leftarrow	60	7	53	589 642.638(20)	-11.4	5.3
78	8	70	\leftarrow	78	7	71	591 273.845(20)	-4.3	-15.1
87	9	79	←	86	10	76	594 696.788(150)	-43.3	-87.2
61	5	57	←	62	2	60	595 211.514(15)	6.2	-3.3
74	21	53	←	75	20	56	596 152.773(40)	15.8	-21.9
69	20	50	\leftarrow	70	19	51	596 992.147(15)	-1.9	-8.7
54	17	37	←	55	16	40	597 049.441(10)	-6.2	0.6
59	18	42	←	60	17	43	597 448.686(15)	-5.3	1.7
87	11	77	←	86	12	74	598 308.577(100)	-31.9	33.3
79	9	71	←	78	10	68	598 780.232(40)	3.9	0.2
64	6	58	←	64	5	59	598 870.113(10)	11.7	-0.9
90	10	80	←	90	9	81	604 987.346(50)	-17.1	-4.8
88	10	78	←	88	9	79	606 878 473(30)	40	16.4
71	6	66	+	72	3	69	608 950 898(15)	-86	_39
53	4	50	+	54	1	53	610 672 068(20)	59	16
86	9	77	_	86	8	78	613 196 700(15)	77	44
78	22	56		70	21	59	613 688 670(100)	23.0	38.6
85	0	77	5	84	10	74	613 832 775(40)	55.0	24.3
05	10	74	-	01	10	74	614 447 625(60)	96	17.1
03	10	74	÷	82	10	71	614 502 717(40)	10.7	17.1
72	31	73	-	30	10	70	615 016 245(50)	10.7	-0.4
/3	21	55	÷	/4	20	J4	C15 (10, 345(30)	-1.1	-23.2
92	10	82	←	92	9	83	615 483.758(80)	-12.5	0.1
50	2	48	←	51	1	51	706 704.602(20)	-12.2	-3.9
67	5	63	←	68	2	66	706 913.649(40)	-5.8	-10.9
/3	22	52	←	74	21	53	707 518.795(60)	-12.2	-6.4
90	9	81	\leftarrow	90	8	82	709 401.348(50)	45.4	35.3
59	4	56	\leftarrow	60	1	59	720 632.005(10)	5.7	4.5
77	6	72	←	78	3	75	722 301.996(60)	4.5	11.2
77	23	55	←	78	22	56	723 997.268(100)	-43.8	3.4
51	3	49	←	52	0	52	725 315.755(15)	-0.1	9.1
67	21	47	←	68	20	48	728 114.583(20)	-24.2	-12.3
92	11	81	\leftarrow	92	10	82	730 959.918(80)	7.4	-30.3
60	3	57	\leftarrow	61	2	60	738 134.248(15)	-0.6	-1.4
77	11	67	\leftarrow	78	8	70	738 260.372(150)	1.8	10.6
90	11	79	←	89	12	78	738 649.459(120)	18.6	62.7
78	5	73	←	79	4	76	739 107.558(30)	30.9	33.5
52	2	50	←	53	1	53	743 042.148(20)	-8.3	2.0
69	5	65	\leftarrow	70	2	68	743 967.302(20)	-4.6	-10.3
71	22	50	←	72	21	51	745 040.392(50)	-9.8	2.3
61	4	58	←	62	1	61	757 135.303(20)	-1.0	-1.6
79	6	74	←	80	3	77	759 786.331(50)	-9.0	-9.3
75	23	53	←	76	22	54	761 344.940(100)	-11.4	43.3
70	4	66	←	71	3	69	761 438.989(30)	-5.5	-12.4
70	22	48	←	71	21	51	763 800.214(100)	10.8	22.8
65	21	45	←	66	20	46	765 791.210(10)	-20.7	-8.8
55					20		,,,,,,,	2017	0.0

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90	11	79	←	90	10	80	766 833.091(50)	9.0	-30.2
92	9	83	←	92	8	84	767 466.649(120)	-62.7	-75.7
62	3	59	←	63	2	62	774 876.880(40)	29.2	28.4
71	5	67	\leftarrow	72	2	70	780 862.523(30)	-9.6	-18.3
64	21	43	\leftarrow	65	20	46	784 626.891(10)	8.4	18.7
63	4	60	←	64	1	63	793 524.340(20)	5.5	4.2
72	4	68	←	73	3	71	798 621.881(40)	-18.2	-29.9
73	23	51	←	74	22	52	798 697.330(150)	56.8	106.5
68	22	46	←	69	21	49	801 319.331(30)	7.8	14.8
63	21	43	←	64	20	44	803 461.207(15)	-15.5	-7.6
20	20	0	←	20	19	1	1 927 842.304(30)	-4.5	4.6
21	20	2	←	21	19	3	1 927 998.798(40)	-42.5	-35.9
22	20	2	←	22	19	3	1 928 161,366(30)	2.5	6.6
23	20	4	←	23	19	5	1928 329.684(30)	7.9	9.9
24	20	4	←	24	19	5	1 928 503 566(40)	-1.1	-1.4
25	20	6	←	25	19	7	1 928 682.827(20)	9.4	6.9
26	20	6	←	26	19	7	1928 867,227(20)	27.6	22.8
27	20	8	←	27	19	9	1 929 056 510(30)	33.2	26.5
29	20	10	←	29	19	11	1929 448.742(20)	11.9	1.8
30	20	10	←	30	19	11	1 929 651 196(40)	4.8	-6.9
31	20	12	←	31	19	13	1929 857.538(25)	19.5	6.3
32	20	12	←	32	19	13	1930.067.441(30)	69	-75
33	20	14	-	33	19	15	1 930 280 606(40)	-45.8	-61.2
34	20	14	E	34	19	15	1930 496 853(40)	-24 7	-40.9
35	20	16	Ļ.	35	19	15	1930 715 784(30)	-25.6	-42.3
36	20	16	-	36	19	17	1 930 937 131(50)	-59	-23.3
37	20	18	-	37	19	19	1 931 160 518(30)	-24.8	_47.2
38	20	18	Ļ.	38	19	19	1931 385 703(40)	24.0	-15.0
50	5	45	Ļ.	49	4	46	1931571663(30)	-29.0	-40.3
39	20	20	-	39	19	21	1 931 612 305(30)	28.1	10.5
40	20	20	, _	40	19	21	1 931 839 941(24)	99	-73
40	20	20	, _	41	19	21	1932 068 330(30)	15.6	_0.9
47	20	22	-	47	19	23	1 932 297 098(20)	28.0	12.2
43	20	22	, _	43	19	25	1932 525 788(50)	-46.2	-61.2
43	20	24	, _	43	19	25	1932 754 238(30)	2.0	-12.0
35	13	23	, _	34	12	23	1932 958 812(30)	-53.7	-44.0
45	20	26	-	45	19	22	1932 981 907(20)	10.7	_2.5
46	20	26	, _	46	19	27	1933 208 376(60)	-53.7	-65.7
28	6	20	, _	27	3	25	1933 324 398(30)	_14 7	-36.8
47	20	22	, _	47	19	29	1933 433 460(20)	17.3	50.0
48	20	20	, _	48	19	29	1933 656 548(20)	13.6	4.0
49	20	30	Ļ.	40	19	31	1933 877 304(40)	65	-21
50	20	30	,	50	10	31	1 934 095 318(50)	0.5	6.7
51	20	32	, _	51	19	33	1934 310 166(60)	-5.5	_11.9
50	20	51	,	58	8	50	1 934 550 591(40)	22.3	77
13	8	36	Ę	43	5	30	1 934 705 394(40)	18.0	20.9
53	20	34	Ę	53	10	35	1 934 705.554(40)	- 18.5	20.5
54	20	34	, _	54	19	35	1934 931 429(40)	13.4	8.4
30	14	16	`	29	13	17	1 935 046 840(20)	_79	_83
41	8	34	-	41	5	37	1 935 117 660(70)	37.2	79.9
55	20	36	-	55	19	37	1935 129 290(40)	43.0	37.6
20	16	1	-	10	15	5	1 935 673 179(30)	10.9	57.0
25	15	11	-	24	13	10	1 935 870 169(20)	27.4	20.7
£5 65	20	46	-	65	19	47	1 936 732 689(100)	_837	_135.4
00	20	-10		05	15	-1/	. 550 / 52.005 (100)	05.7	155.4

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Table 1 (continued)

Transition 1							Frequency, exp. ^a in MHz 2	∆ (calc) ^b in kHz 3	Δ (calc) ^c in kHz 4
39	8	32	Ļ	39	5	35	1 938 137.092(60)	14.8	59.5
75	6	67	ţ	74	8	66	$1 \ 939 \ 206.049(200)$	57.7	78.2
46	11	35	ţ	45	10	36	1940889.752(30)	-1.5	25.8
58	6	49	ţ	57	8	50	1941945.067(40)	25.3	51.0
52	10	42	ţ	51	6	43	1 945 728.205(40)	3.8	33.5

Values in parentheses are experimental uncertainties. The value $\Delta = \nu^{exp-}\nu^{calc}$, where ν^{exp} is taken from Ref. [35], and ν^{calc} is calculated with the parameters from column 2 of Table 2. The value $\Delta = \nu^{exp-}\nu^{calc}$, where ν^{exp} is taken from Ref. [35], and ν^{calc} is calculated with the parameters from column 3 of Table 2.

"allowed" transitions in the parallel bands are

$$\Delta J = 0, \pm 1, \quad \Delta K_a = 0, \quad \Delta K_c = \pm 1.$$

The selection rules for the "allowed" transitions in the perpendicular bands are

 $\Delta J = 0, \pm 1, \quad \Delta K_a = \pm 1, \quad \Delta K_c = \pm 1.$

Besides "allowed" transitions, weak "forbidden" ones are allowed both in the parallel and perpendicular bands. The selection rules have the following form:

$$\Delta J = 0, \pm 1, \quad \Delta K_a = even, \quad \Delta K_c = odd,$$

and

 $\Delta J = 0, \pm 1, \quad \Delta K_a = odd, \quad \Delta K_c = odd.$

In accordance with the symmetry properties, the effective Hamiltonian of the XY₂-type molecule of the $C_{2\nu}$ symmetry has been discussed in the spectroscopic literature many times (see, e.g., Refs. [45-47]). For consistency, we will briefly present it here without detailed

Table 2

Spectroscopic parameters of the ground vibrational state of the SO₂ molecule (in cm⁻¹).^a

Parameter	PS ^b	Ref. [35] ^c
1	2	3
Е		
Ā	2.02735420407(163)	2.02735433
В	0.344173882136(410)	0.3441739084
С	0.293526503766(428)	0.293526529
$\Delta_K \times 10^4$	0.864015421(289)	0.8640369
$\Delta_{JK} \times 10^5$	-0.390123832(126)	-0.3901187
$\Delta_J imes 10^6$	0.220539487(159)	0.220549
$\delta_K imes 10^6$	0.846291509(854)	0.846284
$\delta_J imes 10^7$	0.567423257(229)	0.5674232
$H_K \times 10^7$	0.12360428(155)	0.12375
$H_{KJ} imes 10^9$	-0.64960685(842)	-0.64936
$H_{JK} \times 10^{11}$	0.1160311(386)	0.116
$H_{I} \times 10^{12}$	0.3746170(246)	0.37589
$h_K \times 10^9$	0.5679872(101)	0.5670
$h_{I\!K} imes 10^{12}$	-0.243030(283)	-0.23
$h_{l} \times 10^{12}$	0.18300674(481)	0.1829
$L_{K} \times 10^{11}$	-0.2608919(349)	-0.265
$L_{KKI} \times 10^{12}$	0.1807804(411)	0.180
$L_{lK} \times 10^{13}$	-0.1097261(451)	-0.109
$L_{IJK} \times 10^{17}$	-0.99508(476)	-0.88
$L_{I} \times 10^{17}$	-0.110360(128)	-0.116
$l_K \times 10^{12}$	-0.319192(117)	-0.32
$l_{KJ} imes 10^{14}$	0.254416(212)	0.27
$l_{lK} \times 10^{17}$		-0.2
$l_{I} \times 10^{18}$	-0.607568(294)	-0.597
$P_{K} \times 10^{15}$	0.584163(354)	0.649
$P_{KKI} \times 10^{16}$	-0.408713(675)	-0.394
$P_{KI} \times 10^{18}$	-0.623763(944)	-0.703
$P_{IK} \times 10^{19}$	0.649369(650)	0.778
$S_{\kappa} \times 10^{18}$	-0.088587(130)	-0.12
$S_{KKI} \times 10^{20}$	0.81536(427)	0.70
	. ,	

^a Values in parentheses are 1σ confidence intervals (in last digits).

^b Obtained from the fit in the present study.

^c Recalculated from Table 3 of Ref. [35]. Number of kept digits in the values of parameters corresponds to the number of digits in the initial values from Ref. [35].

explanations:

$$H^{\nu,-r.} = \sum_{\nu,\tilde{\nu}} |\nu\rangle \langle \tilde{\nu} | H_{\nu\tilde{\nu}}, \tag{1}$$

where the summation extends over all (in our case, three) interacting vibrational states: $|1\rangle \equiv (100, A_1)$, $|2\rangle \equiv (020, A_1)$, and $|3\rangle \equiv (001, B_1)$. The diagonal operators H_{vv} describe unperturbed rotational structures of the corresponding vibrational states. The nondiagonal operators $H_{v\tilde{v}}$, $(v \neq \tilde{v})$ describe resonance interactions (Fermi, or Coriolis) between the states $|v\rangle$ and $|\tilde{v}\rangle$. The diagonal block operators have the same form for all the three vibrational states involved (Watson's Hamiltonian in the *A*-reduction and I^r representation):

$$\begin{aligned} H_{\nu\nu} &= E^{\nu} + [A^{\nu} - \frac{1}{2} (B^{\nu} + C^{\nu})]J_{z}^{2} + \frac{1}{2} (B^{\nu} + C^{\nu})J^{2} + \frac{1}{2} (B^{\nu} - C^{\nu})J_{xy}^{2} \\ &- \Delta_{K}^{\nu} J_{z}^{2} - \Delta_{JK}^{\nu} J_{z}^{2} J^{2} - \Delta_{J}^{\nu} J^{4} - \delta_{K}^{\nu} [J_{z}^{2}, J_{xy}^{2}] - 2\delta_{J}^{\nu} J^{2} J_{xy}^{2} \\ &+ H_{K}^{\nu} J_{z}^{6} + H_{KJ}^{\nu} J_{z}^{4} J^{2} + H_{JK}^{\nu} J_{z}^{2} J^{4} + H_{J}^{\nu} J^{6} \\ &+ [J_{xy}^{2}, h_{K}^{\nu} J_{z}^{4} + h_{JK}^{\nu} J^{2} J_{z}^{2} + h_{J}^{\nu} J^{4}] \\ &+ L_{K}^{\nu} J_{z}^{8} + L_{KKJ}^{\nu} J_{z}^{6} J^{2} + L_{JK}^{\nu} J_{z}^{4} J^{4} + L_{KJJ}^{\nu} J_{z}^{2} J^{6} + L_{J}^{\nu} J^{8} \\ &+ [J_{xy}^{2}, l_{K}^{\nu} J_{z}^{6} + l_{KJ}^{\nu} J^{2} J_{z}^{4} + l_{JK}^{\nu} J^{4} J_{z}^{2} + l_{J}^{\nu} J^{6}] \\ &+ \dots + P_{K}^{\nu} J_{z}^{10} + P_{KKKJ}^{\nu} J_{z}^{8} J^{2} + P_{KKJ}^{\nu} J_{z}^{6} J^{4} + P_{JJK}^{\nu} J_{z}^{4} J^{6} \\ &+ S_{KJ}^{\nu} J_{z}^{12} + S_{KKKK}^{\nu} J_{z}^{10} J^{2} + \cdots \end{aligned}$$

where J_{α} ($\alpha = x, y, z$) are the components of the angular momentum operator defined in the molecule-fixed coordinate system; $J_{xy}^2 = J_x^2 - J_y^2$; A^{ν} , B^{ν} , and C^{ν} are the effective rotational constants connected with the vibrational states ν , and the other parameters are the different order centrifugal distortion coefficients.

Table 3

Statistical information for the ν_1 , ν_3 , and $2\nu_2$ bands of SO₂ molecule.

We may distinguish between two types of coupling operators $H^{\nu\tilde{\nu}}$, $(\nu \neq \tilde{\nu})$, corresponding to the two different types of resonance interactions which can occur in XY₂ $(C_{2\nu})$ asymmetric top molecules. If the product $\Gamma = \Gamma^{\nu} \otimes \Gamma^{\tilde{\nu}}$ of the symmetry species of the states ν and $\tilde{\nu}$ is equal to A_1 (*i.e.*, $\Gamma^{\nu} = \Gamma^{\tilde{\nu}}$), then the states ν and $\tilde{\nu}$ are connected by an anharmonic Fermi resonance interaction, and the corresponding interaction operator has the form

$$\begin{aligned} H_{\nu\tilde{\nu}} &= {}^{\nu\tilde{\nu}}F_{0} + {}^{\nu\tilde{\nu}}F_{K}J_{z}^{2} + {}^{\nu\tilde{\nu}}F_{J}J^{2} + {}^{\nu\tilde{\nu}}F_{KK}J_{z}^{4} + {}^{\nu\tilde{\nu}}F_{KJ}J_{z}^{2}J^{2} \\ &+ {}^{\nu\tilde{\nu}}F_{JJ}J^{4} + \dots + {}^{\nu\tilde{\nu}}F_{xy}(J_{x}^{2}-J_{y}^{2}) + {}^{\nu\tilde{\nu}}F_{Kxy}\{J_{z}^{2}, (J_{x}^{2}-J_{y}^{2})\}_{+} \\ &+ 2{}^{\nu\tilde{\nu}}F_{Ixy}J^{2}(J_{x}^{2}-J_{y}^{2}) + \dots \end{aligned}$$
(3)

If the product is $\Gamma = B_1$, then the states ν and $\tilde{\nu}$ are connected by a Coriolis resonance interaction of the form

$$\begin{aligned} H_{\nu\bar{\nu}} &= i J_{\nu} H_{\nu\bar{\nu}}^{(1)} + H_{\nu\bar{\nu}}^{(1)} i J_{\nu} + \{J_{x}, J_{z}\}_{+} H_{\nu\bar{\nu}}^{(2)} + H_{\nu\bar{\nu}}^{(2)} \{J_{x}, J_{z}\}_{+} \\ &+ \{i J_{\nu}, (J_{x}^{2} - J_{\nu}^{2})\}_{+} H_{\nu\bar{\nu}}^{(3)} + H_{\nu\bar{\nu}}^{(3)} \{i J_{\nu}, (J_{x}^{2} - J_{\nu}^{2})\}_{+} + \cdots$$

$$\end{aligned}$$

The operators $H_{v\tilde{v}}^{(i)}$, i = 1, 2, 3, ... in Eq. (4) have the form

$$\begin{aligned} H^{(i)}_{\nu\bar{\nu}} &= \frac{1}{2} {}^{\nu\bar{\nu}} C^{i}_{I} + {}^{\nu\bar{\nu}} C^{i}_{K} J^{2}_{z} + \frac{1}{2} {}^{\nu\bar{\nu}} C^{i}_{J} J^{2} + {}^{\nu\bar{\nu}} C^{i}_{KK} J^{4}_{z} + {}^{\nu\bar{\nu}} C^{i}_{KJ} J^{2}_{z} J^{2} \\ &+ \frac{1}{2} {}^{\nu\bar{\nu}} C^{i}_{JJ} J^{4}_{z} + {}^{\nu\bar{\nu}} C^{i}_{KKK} J^{4}_{z} + {}^{\nu\bar{\nu}} C^{i}_{KKJ} J^{4}_{z} J^{2} \\ &+ {}^{\nu\bar{\nu}} C^{i}_{KIJ} J^{2}_{z} J^{4} + \frac{1}{2} {}^{\nu\bar{\nu}} C^{i}_{IIJ} J^{6}_{z} + \cdots \end{aligned}$$

$$(5)$$

To prevent confusion in the label notations used, we should mention that the axis of symmetry of the XY_2 molecule in our case is not the *z*-axis, but the *x*-axis. This fact is the consequence of using the diagonal blocks (2) of the Hamiltonian (1) in the form of Watson's operator in *A*-reduction and *I*^{*r*}-representation. However, for the point group symmetry assignment of normal modes we use the

Band	Center in cm ⁻¹	J ^{max}	K _a ^{max}	Nl ^a	rms_l in 10^{-5} cm ⁻¹	m _{l1} b in %	m _{l2} b in %	m _{l3} b in %	N _t ^c	rms_t in 10^{-5} cm ⁻¹	m _{t1} ^b in %	m _{t2} ^b in %	m _{t3} b in %
1	2	3	4	5	6	7	8	9	10	11	12	13	14
Ground With param. from Ref. [35] ^d		92	23						149	28.2 ^d			
With our param. ^d With param. from Ref. [35] ^e		92 53	23 34						149 78	22.5 ^d 269.1	3.8	11.5	84.7
With our param. ^e		53	34						78	10.8	73.1	20.5	6.4
ν_1 Ref. [26] ($J \le 69/K_q \le 28$)	1151.71295	69	28	1216	21.0		70.7	29.3					
Our $(J \le 69/K_a \le 28)$ Our (total)		69 89	28 37	1562 1913	3.9 6.1	97.3 93.3	2.4 5.1	0.3 1.6	6925 7618	12.2 12.7	80.4 78.4	9.4 10.6	10.2 11.0
ν_3 Ref. [26] ($J \le 89/K_a \le 24$)	1362.06030	89	24	884	21.0		70.7	29.3					
Our $(J \le 89/K_a \le 24)$ Our (total)		89 109	24 28	1653 1838	8.2 9.7	87.7 84.1	9.2 10.6	3.1 5.3	3695 3952	15.0 15.6	69.9 68.3	14.3 14.7	15.8 17.0
2ν ₂ Our	1035.12639	54	9	312	13.9	68.0	20.5	11.5	561	21.1	43.6	25.1	31.3

^a N_l is the numbers of upper levels.

^b Here $m_i = n_i/N \times 100\%$ (*i*=1, 2, 3); n_1, n_2 , and n_3 are the numbers of levels (transitions) for which the differences $\delta = |E^{exp} - E^{calc.}|$ (for columns 7–9), or $\delta = |\nu^{exp} - \nu^{calc}|$ (for columns 12–14) satisfy the conditions $\delta \le 10 \times 10^{-5}$ cm⁻¹, 10×10^{-5} cm⁻¹ < $\delta \le 20 \times 10^{-5}$ cm⁻¹, and $\delta > 20 \times 10^{-5}$ cm⁻¹.

^c N_t is the numbers of transitions.

^d Reproduction of sub-millimeter wave transitions from Ref. [35]: $K_a \leq 23$. The *rms* in column 11 is in kHz.

^e Reproduction of Ground State Combination Differences obtained from analysis of our FTIR experimental data: $29 \le K_a \le 34$.

standard convention with $z=C_2$. For further, more detailed discussion of this type of effective Hamiltonian model we refer to [45].

4. Analysis of spectra and re-determination of ground state spectroscopic parameters

Survey weak spectrum of the ν_1 and ν_3 bands of the SO₂ molecule is shown on the upper part of Fig. 1. Fig. 2 presents the S-spectrum of the ν_1 band. In this case, the P-branch of very weak $2\nu_2$ band is seen in the lower wavenumber region. For illustration of quality of the experimentally recorded spectra, upper part of Fig. 3 presents small portion of the high resolution spectrum in the R-branch of the ν_1 band. As the analysis showed, in both the ν_1 and ν_3 , and $2\nu_2$ bands, at least, two of three (P-, Q-, and/or R-) branches are clearly pronounced. As a consequence, one can expect a good fulfillment of the "ground state combination differences" principle.

In the analysis of the recorded spectra we were able to assign transitions with the maximum values of quantum numbers $J^{max.} = 109$ and $K_a^{max.} = 37$. Correctness of assignments was controlled by the construction of corresponding "experimental" Ground State Combination Differences. It was found that, starting with the value of quantum number $K_a = 26-27$, difference between GSCD values obtained from experimental transitions and calculated ones obtained by the best set of the ground state rotational parameters, Ref. [35], starts increasing quickly with increase of quantum number K_q . In this case, differences between "experimental" and "calculated" GSCD values were increased from about $1-1.5 \times 10^{-4} \text{ cm}^{-1}$ for $K_a \leq 25$ up to 60.5×10^{-4} cm⁻¹ for $K_a = 34$. This circumstance becomes clear if one takes into account that in [35] the ground state rotational parameters were obtained on the basis of high-accurate experimental sub-millimeter wave transitions with the maximum value of quantum number $K_a^{max.} = 23$. For this reason, in the present paper we re-analyze the rotational structure of the SO₂ ground vibrational state on the basis of the ground state

Table 4

Spectroscopic parameters of the (100), (001), and (020) vibrational states of the SO₂ molecule (in cm⁻¹).^a

Parameter 1	(100), our 2	(100) from [26] 3	(001), our 4	(001) from [26] 5	(020), our 6	(020) from [26] 7
$\begin{array}{c} \text{Parameter } 1 \\ \hline \\ E \\ A \\ B \\ C \\ \Delta_{JK} \times 10^{4} \\ \Delta_{JK} \times 10^{5} \\ \Delta_{J} \times 10^{6} \\ \delta_{V} \times 10^{7} \\ H_{K} \times 10^{7} \\ H_{KJ} \times 10^{9} \\ H_{JK} \times 10^{11} \\ H_{J} \times 10^{12} \\ h_{K} \times 10^{9} \\ h_{JK} \times 10^{12} \\ L_{JK} \times 10^{12} \\ L_{JK} \times 10^{12} \\ L_{JK} \times 10^{13} \\ L_{JK} \times 10^{16} \\ L_{J} \times 10^{15} \\ P_{KKJ} \times 10^{16} \\ P_{KJ} \times 10^{16} \\ P_{KJ} \times 10^{18} \end{array}$	(100), our 2 1151.68575415 (466) 2.0284219524 (969) 0.3425072831 (280) 0.2922516577 (118) 0.87785264 (873) -0.3905198 (726) 0.2193355 (109) 0.846291509 0.5676407 (604) 0.1272614 (117) -0.656738 (253) 0.1160311 0.371073 (328) 0.5679872 -0.243030 0.184502 (319) -0.271360 (112) 0.1868310(860) -0.110391 (183) -0.099508 -0.110360 -0.324991 (886) 0.284853 (540) -0.607568 0.599887 (390) -0.408713 -0.623763	(100) from [26] 3 1151.712950 2.028436263 0.3425122666 0.2921140821 0.87644839 -0.37655179 0.21977784 0.922469 0.5653892 0.1283702 -0.715856 0.1325 0.38265 0.553916 -0.37865 0.178527 -0.280590 0.125454 0.007728 -1.2014 -0.3099 0.68478 -0.26122	(001), our 4 1362.06030242(716) 2.006648486 (197) 0.3430035682 (288) 0.2922918154 (122) 0.8487019 (198) -0.3921223 (802) 0.2234330 (110) 0.8542019 (820) 0.5719439 (545) 0.1202984 (677) -0.651555 (295) 0.1160311 0.383059 (564) 0.577289 (542) -0.243030 0.184898 (178) -0.24940 (108) 0.175030 (216) -0.1097261 -0.099508 -0.10166 (312) -0.315084(552) 0.254416 -0.607568 0.55386 (604) -0.408713 -0.623763	(001) from [26] 5 1362.060336 2.006644631 0.3430079384 0.2924294870 0.8503404 -0.4069928 0.22308590 0.7891669 0.5743462 0.120473 -0.608500 0.94826 0.37734 0.90549 -0.37865 0.190584 -0.26921 0.16636 0.007728 -1.8914 -0.2910 0.7101 -0.26122	(020), our 6 1035.1535902 (118) 2.107666447 (651) 0.3443180685 (580) 0.2924639658 (337) 1.0631021 (904) -0.4255028 (501) 0.2216133 (260) 1.23853 (158) 0.574481 (140) 0.182918 -0.8644971 0.79840 0.373265 0.954963 -1.3097 0.183500 -0.478860 0.29657 -0.1993 -0.7190 -0.0957 -0.5895 0.273 -0.24 -0.69185 1.443 -0.78360 -1.286	(020) from [26] 7 1035.126371 2.107649092 0.3443172408 0.2924635974 1.0638447 -0.4257225 0.2212268 1.236181 0.5742303 0.188955 -0.86163 0.554 0.26162 0.927 -2.78 0.194 -0.437 0.158
$P_{JK} imes 10^{19} \ S_K imes 10^{19} \ S_{KKJ} imes 10^{20}$	0.649369 -0.88587 0.81536		0.649369 0.88587 0.81536		1.666 -3.30 1.615	

^a Values in parentheses are 1σ standard errors. Parameters presented without standard errors have been fixed to the values of the ground vibrational state ((100) and (001) states) from column 2 of Table 2, or to the values ((020) state) estimated in accordance with the formula (6).

combination differences determined in the frame of analysis of our FTIR experimental data (in this case, 78 FTIR combination differences with the values $J^{max.} = 53$ and $K_a^{max.} = 34$ were used). The 149 sub-millimeter wave transitions from Ref. [35] (they are reproduced in column 2 of Table 1) were also added to the set of initial data used in the fit of the ground state spectroscopic parameters. The values of parameters obtained from the weighted fit are presented in column 2 of Table 2. Values in parenthesis are 1σ statistical confidence intervals. For comparison, column 3 of Table 2 presents the values of the ground

Table 5

Resonance interaction parameters for the (100), (020), and (001) vibrational states of the SO_2 molecule (in $cm^{-1})^{\rm .a}$

state spectroscopic parameters that are reproduced from

Parameter	Value
(100),(020) F ₀	1.78
(100),(001) C ¹	-0.17
$^{(100),(001)}C^3 \times 10^6$	0.31773(644)
$^{((100),(020)}F_{Kxy} \times 10^7$	0.714042(816)
$^{(100),(001)}C_{k}^{1} \times 10^{4}$	-0.16849(214)
$(100),(001)C_{l}^{3} \times 10^{11}$	-0.2335(125)
$(100),(001)C^2 \times 10^2$	0.20827(167)

^a Values in parentheses are 1σ standard errors. Parameters presented without standard errors have been estimated on the basis of the potential energy surface parameters of SO₂ from Ref. [48] and were constrained in the fit.

Table 6

Heterodyne frequencies of SO₂ in the ν_3 band.

Ref. [35]. One can see good correlation between both sets of parameters. At the same time our new set of parameters not only reproduce better the FTIR Ground State Combination Differences (the *rms* deviation of 78 GSCD, $29 \ge K_a \ge 34$, obtained from our experimental data, being calculated with the parameters from Ref. [35] is 269.1×10^{-5} cm⁻¹; being calculated with our set of parameters, that value is decreased up to 10.8×10^{-5} cm⁻¹, see statistical Table 3 for more details), but also reproduces with the better rms-deviation values the sub-millimeter wave line positions from Ref. [35] (compare columns 3 and 4 of Table 1; the *rms* deviation is 28.2 kHz and 22.5 kHz at calculation with parameters from [35] and our parameters from column 2 of Table 2, respectively). The number of fitted parameters in our case is even less than in [35].

5. Re-analysis of the rotational structures of the vibrational states (100), (001), and (020)

The new ground state parameters obtained in Section 4 were used then for calculation of ground state rotational energies which, in turn, were used in the re-assignments of transitions in the recorded FTIR spectra. The list of more than 12 130 finally assigned transitions is presented in the Supplementary Materials (see also the statistical information in Table 3). From these transitions we obtained 4063 upper ro-vibrational energies (1913, 1838, and 312 upper energies for the states (100), (001), and (020), respectively) which were used then as input data in a weighted least square fit with the aim to determine rotational, centrifugal distortion, and resonance interaction parameters of the

Transition 1	Frequency, exp.ª in MHz 2	∆ <i>I</i> , ^b in MHz 3	∆ <i>II</i> , ^c in MHz 4	Transition 1	Frequency, exp.ª in MHz 2	∆ <i>I</i> , ^b in MHz 3	∆ <i>II</i> , ^c in MHz 4
${}^{q}\mathrm{P}_{6}(18)$	40 457 019.8(5.0)	1.5	-0.3	${}^{q}\mathrm{R}_{1}(13)$	41 079 955.2(5.0)	1.5	0.3
$^{q}P_{0}(20)$	40 458 986.4(5.0)	1.4	0.0	$^{q}R_{5}(13)$	41 080 193.8(3.0)	-3.1	-4.6
$^{q}P_{10}(16)$	40 460 775.5(9.0)	3.5	2.0	${}^{q}R_{10}(26)$	41 267 578.4(5.0)	-1.0	-1.9
${}^{q}P_{8}(17)$	40 461 362.9(5.0)	0.3	-1.2	${}^{q}R_{6}(24)$	41 271 076.8(4.0)	0.6	-0.9
$^{q}P_{2}(19)$	40 462 193.6(5.0)	2.1	0.6	${}^{q}R_{8}(25)$	41 271 318.4(4.0)	-0.7	-1.9
${}^{q}\mathrm{P}_{5}(18)$	40 463 046.5(5.0)	2.1	0.6	${}^{q}R_{11}(27)$	41 272 779.1(4.0)	-1.4	-1.7
${}^{q}\mathrm{P}_{2}(18)$	40 464 934.6(5.0)	1.5	0.3	${}^{q}R_{1}(25)$	41 275 291.4(4.0)	1.3	-0.2
$^{q}Q_{12}(21)$	40 734 536.6(4.0)	0.4	-0.5	${}^{q}R_{8}(38)$	41 494 155.7(6.0)	1.2	0.6
${}^{q}P_{2}(5)$	40 734 999.1(4.0)	-0.8	-1.7	$^{q}R_{12}(41)$	41 495 464.1(8.0)	-0.8	-0.5
$^{q}Q_{12}(20)$	40 735 926.4(4.0)	-0.5	-1.7	${}^{q}R_{6}(37)$	41 496 666.6(4.0)	1.0	-0.2
$^{q}Q_{12}(19)$	40 737 249.7(4.0)	-1.7	-2.9	${}^{q}R_{4}(37)$	41 499 389.5(4.0)	0.0	-1.8
$^{q}Q_{12}(18)$	40 738 508.3(4.0)	-1.3	-2.5	${}^{q}R_{3}(36)$	41 500 253.4(4.0)	1.1	-0.7
${}^{q}\mathrm{P}_{1}(5)$	40 740 637.5(4.0)	1.8	0.9	${}^{q}R_{5}(37)$	41 502 232.2(4.0)	1.0	-0.5
$^{q}Q_{12}(16)$	40 740 829.0(4.0)	1.4	0.2	${}^{q}R_{0}(40)$	41 504 180.9(6.0)	2.2	1.6
$^{q}Q_{11}(25)$	40 741 496.8(4.0)	1.3	0.7	${}^{q}R_{7}(38)$	41 504 742.1(5.0)	0.1	-0.8
$^{q}Q_{12}(15)$	40 741 887.7(4.0)	0.6	-0.9	${}^{q}R_{10}(40)$	41 504 915.0(6.0)	-1.1	-1.1
$^{q}Q_{12}(13)$	40 743 809.8(6.0)	2.3	0.8	$^{q}R_{2}(38)$	41 505 369.2(4.0)	1.3	-0.8
${}^{q}R_{0}(2)$	40 890 362.9(6.0)	13.1	12.0	${}^{q}R_{8}(56)$	41 792 579.5(4.0)	-1.0	-0.1
${}^{q}R_{3}(7)$	40 979 127.8(20.0)	5.7	4.5	$^{q}R_{10}(58)$	41 793 377.8(5.0)	-1.9	-5.8
${}^{q}R_{6}(8)$	40 981 712.6(10.0)	-1.0	-2.4	$^{q}R_{11}(59)$	41794759.0(4.0)	-2.5	-1.6
${}^{q}R_{8}(9)$	40 983 895.9(9.0)	2.6	1.1	$^{q}R_{3}(58)$	41 796 023.8(6.0)	-3.7	-3.7
${}^{q}R_{5}(8)$	40 988 165.8(15.0)	-4.4	-5.9	${}^{q}R_{1}(60)$	41 796 222.6(5.0)	0.5	2.9
${}^{q}R_{8}(14)$	41 075 376.3(3.0)	1.3	-0.2	${}^{q}R_{5}(57)$	41 796 860.0(5.0)	0.0	-0.6
${}^{q}R_{11}(16)$	41 078 597.4(6.0)	2.6	1.1				

^a Values in parentheses are experimental uncertainties (in MHz).

^b The value $\Delta I = \nu^{exp.} - \nu^{calc.}$, where $\nu^{exp.}$ is taken from Ref. [23], and $\nu^{calc.}$ is calculated with the parameters from column 2 of Table 2, columns 2, 4, 6 of Table 4 and 5.

^c The values $\Delta II = \nu^{exp.} - \nu^{calc.}$ are reproduced from the last column of Table V, Ref. [26].

Table 7					
Heterodyne	frequencies	of SO_2	in	the ν_1	band.

Transition	Wavenumber, exp. ^a in cm ⁻¹	$\Delta I^{\mathbf{b}}$	ΔII^{c}	Transition	Wavenumber, exp. ^a in cm ⁻¹	$\Delta I^{\rm b}$	ΔII^{c}
1	2	3	4	1	2	3	4
${}^{p}P_{10}(48)$	1085.8791(2)	3.0	2.5	${}^{p}P_{13}(43)$	1079.6768(2)	0.7	0.3
${}^{p}P_{13}(35)$	1085.7302(2)	-0.6	-0.8	${}^{p}\mathrm{P}_{17}(26)$	1079.6722(2)	0.9	0.0
${}^{p}P_{16}(22)$	1085.6640(2)	-0.4	-1.2	${}^{p}P_{18}(23)$	1078.7424(2)	-1.6	-2.7
${}^{p}P_{11}(45)$	1084.7518(2)	1.5	1.2	${}^{p}P_{14}(40)$	1078.7327(2)	-3.4	-3.8
${}^{p}P_{14}(32)$	1084.7412(2)	1.0	0.7	${}^{p}P_{15}(36)$	1078.5654(2)	-0.9	-1.4
${}^{p}P_{17}(19)$	1084.6661(2)	0.2	-0.9	${}^{p}P_{19}(19)$	1078.5249(2)	1.5	0.2
${}^{p}P_{15}(28)$	1084.4941(2)	0.6	0.1	${}^{p}P_{16}(32)$	1078.3954(2)	0.3	-0.3
${}^{p}P_{10}(51)$	1083.6276(2)	0.9	0.3	${}^{p}P_{17}(29)$	1077.4839(2)	2.4	1.6
${}^{p}Q_{21}(31)$	1083.5631(2)	0.3	1.5	${}^{p}P_{13}(46)$	1077.3667(2)	0.7	0.1
${}^{p}P_{16}(25)$	1083.5164(2)	-0.8	-1.5	${}^{p}P_{25}(18)$	1077.3043(2)	0.2	-0.8
${}^{p}P_{13}(38)$	1083.4793(2)	0.5	0.2	${}^{p}P_{12}(52)$	1076.0146(2)	-1.7	-2.6
${}^{p}Q_{21}(32)$	1083.4529(2)	1.5	2.8	${}^{p}P_{17}(31)$	1076.0089(2)	-0.1	-0.9
${}^{p}P_{9}(56)$	1083.4368(2)	0.1	-0.4	${}^{p}P_{18}(27)$	1075.8530(2)	-0.2	-1.1
${}^{p}Q_{21}(33)$	1083.3391(2)	0.2	1.6	${}^{p}P_{20}(20)$	1074.7958(2)	1.1	0.2
${}^{p}P_{14}(34)$	1083.2556(2)	-3.0	-3.3	${}^{p}P_{15}(41)$	1074.7655(2)	-1.1	-1.8
${}^{p}P_{17}(22)$	1082.5452(2)	-0.6	-1.6	${}^{p}P_{16}(37)$	1074.6457(2)	0.4	-0.3
${}^{p}P_{14}(35)$	1082.5091(2)	0.9	0.5	${}^{p}P_{17}(33)$	1074.5217(2)	0.1	-0.6
${}^{p}P_{11}(48)$	1082.4653(2)	-1.0	-1.6	${}^{p}P_{19}(28)$	1072.0598(2)	-0.3	-1.2
${}^{p}P_{15}(31)$	1082.2932(2)	-0.5	-1.0	${}^{p}P_{20}(24)$	1071.9389(2)	1.0	0.3
${}^{p}P_{18}(18)$	1082.2808(2)	-1.0	-2.3	${}^{p}P_{14}(49)$	1071.7722(2)	-2.0	-2.8
${}^{p}P_{12}(44)$	1082.1895(2)	0.1	-0.3	${}^{p}P_{18}(34)$	1070.6730(2)	1.0	0.2
${}^{p}P_{16}(27)$	1082.0690(2)	0.9	0.3	${}^{p}P_{20}(26)$	1070.4904(2)	-0.1	-0.6
${}^{p}P_{16}(28)$	1081.3403(2)	-1.0	-1.7	${}^{p}P_{22}(21)$	1070.3808(2)	-0.1	0.3
${}^{p}P_{17}(24)$	1081.1152(2)	0.7	-0.2	${}^{p}P_{18}(36)$	1069.1647(2)	1.3	0.5
^{<i>p</i>} P ₁₄ (37)	1081.0069(2)	0.5	0.2	${}^{p}P_{19}(32)$	1069.1017(2)	-0.3	-1.0
${}^{p}\mathrm{P}_{18}(20)$	1080.8756(2)	1.5	0.2	${}^{p}P_{20}(28)$	1069.0289(2)	0.9	0.5
${}^{p}P_{12}(47)$	1079.8872(2)	-0.2	-0.4	${}^{p}P_{21}(24)$	1068.9440(2)	0.6	1.1
${}^{p}P_{16}(30)$	1079.8739(2)	-1.6	-2.2				

^a Values in parentheses in column 2 are experimental uncertainties (in 10^{-4} cm⁻¹).

^b The value $\Delta I = \nu^{exp.} - \nu^{calc.}$ (in 10⁻⁴ cm⁻¹), where $\nu^{exp.}$ is taken from Refs. [12,26], and $\nu^{calc.}$ is calculated with the parameters from column 2 of Table 2, columns 2, 4, 6 of Tables 4 and 5.

^c The values $\Delta II = \nu^{exp.} - \nu^{calc.}$ (in 10⁻⁴ cm⁻¹) are reproduced from the before last column of Table VI, Ref. [26].

states (100), (001), and (020). Numerous resonance interactions between all three vibrational states were taken into account.

The fit was made with the Hamiltonian model discussed in Section 3. The initial values of rotational and centrifugal distortion parameters of the (100) and (001) states were taken to be equal to the values of corresponding parameters of the ground vibrational state. The initial values of spectroscopic parameters of the (020) state were estimated in accordance with the formula

$$P^{(020)} = P^{(000)} + 2(P^{(010)} - P^{(000)}), \tag{6}$$

where $P^{(000)}$ or $P^{(010)}$ is any of rotational or centrifugal distortion parameter of the ground or (010) vibrational state from Ref. [35], respectively; $P^{(020)}$ is a corresponding estimated parameter of the (020) vibrational state. The values of two main resonance interaction parameters, $^{(100),(020)}F_0$ and $^{(100),(001)}C^1$, have been estimated on the basis of the potential energy surface parameters of SO₂ from Ref. [48] and were constrained in the fit. The reason for the last was in the following. It is known (see, e.g., Refs. [49,50]) that the main both Fermi- and Coriolis-interaction parameters are strongly correlated with pure vibrational energies, E^ν , or effective rotational parameters, A^ν , B^ν , C^ν , respectively. In our case, when we tried to vary the $^{(100),(020)}F_0$ and/or $^{(100),(001)}C^1$ parameters, their absolute values change more than twice that is absolutely unsuitable from the physical point of view.

For this reason we preferred to fix values of these parameters to theoretically estimated ones.

Results of the fit with the Hamiltonian, Eqs. (1)–(5), are presented in columns 2, 4, and 6 of Table 4 and in Table 5 (values in parentheses are 1σ statistical confidence intervals). Parameters presented without confidence intervals have been constrained to their initial values (see above). To take into account numerous accidental resonance interactions between all three studied vibrational states, (100), (001), and (020) (especially for high values of quantum numbers J and K_a), we introduced into consideration some resonance interaction parameters. They are presented in Table 5 together with their 1σ confidence intervals (in parentheses). In this case, values of the main resonance interaction parameters, ${}^{(100),(020)}F_0$ and ${}^{(100),(001)}C^1$, were constrained to their predicted ones (see above). To compare results of the present study with the previous ones, columns 3, 5, and 7 of Table 4 show the values of spectroscopic parameters from earlier study, Ref. [26]. One can see good correlation between both sets of parameters. At the same time our parameters reproduce better our FTIR experimental data. In particular, the 4063 ro-vibrational energies obtained from our experimental data are reproduced by our set of parameters with the $d_{rms} = 8.7 \times 10^{-5}$ cm⁻¹. For comparison, the same 4063 rovibrational energies are reproduced by the parameters from Ref. [26] with the $d_{rms} = 3329.5 \times 10^{-5} \text{ cm}^{-1}$ (see

Table 8MW frequencies of SO2 in the ν_1 band.

Tra 1	nsiti	on					Frequency, exp. in MHZ 2	⊿, ^a in MHZ 3
1	1	1	←	0	0	0	69 566.06	0.09
1	1	1	←	2	0	2	12 522.89	0.17
2	1	1	←	2	0	2	53 595.80	0.10
3	2	2	←	4	1	3	70 293.43	0.04
4	1	3	←	4	0	4	59 260.96	-0.21
4	2	2	\leftarrow	5	1	5	70 765.92	0.09
5	2	4	\leftarrow	6	1	5	24 275.46	0.10
6	1	5	\leftarrow	6	0	6	68 951.16	-0.69
6	2	4	\leftarrow	7	1	7	44 794.50	0.02
8	1	7	\leftarrow	7	2	6	24 301.59	0.35
8	2	6	\leftarrow	9	1	9	24 888.00	0.04
10	2	8	\leftarrow	11	1	11	12 597.69	0.14
10	3	7	\leftarrow	11	2	10	53 378.90	-0.29
12	2	10	\leftarrow	13	1	13	9172.50	0.26
12	3	9	\leftarrow	13	2	12	21 768.64	0.20
14	2	12	\leftarrow	13	3	11	45 911.10	-0.11
14	2	12	\leftarrow	15	1	15	15 230.50	0.14
15	4	12	\leftarrow	16	3	13	40 720.80	-0.07
16	2	14	\leftarrow	17	1	17	30 700.50	-0.05
17	2	16	\leftarrow	16	3	13	27 386.15	-0.07
16	4	12	\leftarrow	17	3	15	33 996.10	0.05
19	2	18	\leftarrow	18	3	15	41 642.40	0.03
21	2	20	\leftarrow	20	3	17	46 929.50	0.38
21	3	19	\leftarrow	20	4	16	35 084.30	-0.02
21	5	17	\leftarrow	22	4	18	26 977.80	-0.04
22	5	17	\leftarrow	23	4	20	14 974.30	0.03
24	4	20	\leftarrow	23	5	19	19 226.75	-0.05
25	4	22	\leftarrow	24	5	19	23 788.00	0.04
26	6	20	\leftarrow	27	5	23	37 515.60	-0.07
27	6	22	\leftarrow	28	5	23	12 563.00	-0.04
30	5	25	\leftarrow	29	6	24	32 381.80	-0.13
31	5	27	\leftarrow	30	6	24	43 878.40	-0.18
31	7	25	\leftarrow	32	6	26	40 163.00	0.01
32	7	25	\leftarrow	33	6	28	20 835.10	-0.03
35	6	30	\leftarrow	34	7	27	20 689.80	-0.02
36	8	28	\leftarrow	37	7	31	45 834.00	-0.16
37	8	30	\leftarrow	38	7	31	24 431.80	-0.15
40	7	33	\leftarrow	39	8	32	17 994.70	0.23
41	7	35	\leftarrow	40	8	32	37 593.40	0.27
42	9	33	\leftarrow	43	8	36	29 668.00	-0.12
46	8	38	\leftarrow	45	9	37	33 733.40	0.09

^a The value $\Delta = \nu^{exp.} - \nu^{calc.}$, where $\nu^{exp.}$ is taken from Ref. [26], and $\nu^{calc.}$ is calculated with the parameters from column 2 of Table 2, columns 2, 4, 6 of Tables 4 and 5.

also statistical information in Table 3). In this case, differences between experimental energy values and ones calculated with the parameters from Ref. [26] are increased quickly with the increase of K_a quantum number, and for $K_a=37$ the difference reaches the value of 0.62 cm⁻¹.

As one more illustration of the correctness of our results, we send the reader to Tables 6–8. Columns 2 of these tables reproduce high accurate heterodyne experimental line positions of the ν_3 band from Ref. [23] (Table 6), ν_1 band from Ref. [12] (Table 7), and microwave rotational transitions in the excited (100) vibrational state (Table 8) from Ref. [26], respectively. Columns 3 of these three tables present the values of differences Δ between experimental line positions and corresponding values calculated with our parameters. One can see that values of Δ are not worse than the experimental uncertainties (the lasts are given in

parentheses in column 2 of Tables 6 and 7). Columns 4 of Tables 6 and 7 present, for comparison, corresponding Δ -values from Ref. [26]. It is important to note that the experimental data from columns 2 of Tables 6–8 were not used in our fit as an input data. It means that one can consider results shown in the columns 3 of Tables 6–8, as a prediction. At the same time, our predicted Δ -values are, at least, not worse than ones from Ref. [26] (compare columns 3 and 4 of Tables 6 and 7) in spite of the fact that both the heterodyne and MW data reproduced in column 2 of Tables 6–8 were used in the fit of Ref. [26].

6. Conclusion

We re-analyzed the high resolution ro-vibrational structures of the ν_1 and ν_3 vibrational bands and, as a result, assigned about two times more transitions than was made before. Moreover, the weak $2\nu_2$ band has been observed and analyzed for the first time. We determined parameters of the Hamiltonian that reproduce the initial "experimental" ro-vibrational energies of the states (100), (001), and (020) with the *rms* deviation of 6.1×10^{-5} cm⁻¹, 9.7×10^{-5} cm⁻¹, and 13.9×10^{-5} cm⁻¹, respectively, that is close to uncertainties in experimental line positions.

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Appendix A. Supplementary data

Supplementary data associated with this article can be found in the online version at http://dx.doi.org.10.1016/j. jqsrt.2013.04.011.

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