## **Tomsk Polytechnic University**

# PHYSICS II Reports on Laboratory Experiments

UDC 53 (076)

PHYSICS II. Electricity and Magnetism. Electromagnetic Oscillations and Waves. Reports on Laboratory Experiments. Tomsk: TPU Press, 2002, 31pp.

These instructions have been approved by the Department of Theoretical and Experimental Physics and the Department of General Physics of TPU

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All the laboratory experiments must be performed in the TPU Physics Laboratory. The estimation of students' laboratory activity is based on their ability to perform the laboratory experiment and to present the results in conventional format. The report must be carefully prepared. It must include all the measurements and calculations.

#### List of laboratory experiments

- 1. Studying of an Electric Field
- 2. Determination of the Ionic Charge of Hydrogen
- 3. Measurement of Capacitance and Permittivity by Bridge Method
- 4. Determination of High Value Resistance and Capacitance with the Help of Relaxed Oscillations
- 5. Study of Semiconductor Resistance-Temperature Dependence. Determination of Activation Energy.
- 6. Measurement of a Solenoid Magnetic Field Strength.

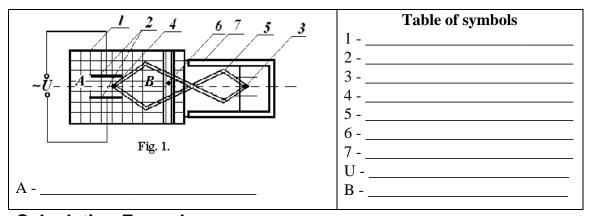
## **Report on Laboratory Experiment No 1**

## **Electric Field Research**

	The student:
Group	
First name	
Last name	
is <b>allowed</b> to do the laboratory	work.
Date	Signature of the teacher
	entials of electric field equipotential surfaces produced by s. Graphically plot electric field strengths on the basis of
<b>Devices and Instrumer</b> Galvanometer, measuring circuforms, plotting paper.	nts uit, electric bath, electrodes of the plane and cylindrical
Theoretical Contents. SElectric field is	Summary
Field strength is	
Electric field potential is	
Equipotantial surface is	

Electric field strength and potential relation	
Method of electric field research	
Principle of determination of the equipotential of	nurfa a a s
Principle of determination of the equipotential s	surraces
Floatria Cabarra	
Electric Scheme	Table of symbols
	$E_2$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$E_6$
	$R_1$
	$R_2$
E <sub>2</sub> , E <sub>2</sub> E <sub>6</sub> E <sub>4</sub>	G
R <sub>4</sub>	U C
D	Tr

## Scheme of the Electrode's Allocation in the Electrolytic Bath.



### **Calculation Formulas**

1. Potential of the equipotential surfaces produced by plate electrodes

(n :	=
------	---

where

$$U-$$

$$x-$$

2. Potential of the equipotential surfaces produced by cylindrical electrodes

where

$$U-$$

$$x-$$

### **Measurement Results**

Necessary data for calculation of potential are determined from graphics of equipotential surfaces. Results of measurements and calculations are recorded in tables.

1. Plate electrodes.

Table 1

Order of	Dista	ances	U	$\varphi$
equipotential surfaces	X	$\ell$	(Volt)	(Volt)
surfaces				

### 2. Cylindrical electrodes

Table 2

Order of	Dista	ances	U	$\varphi$
equipotential	x	d	(Volt)	(Volt)
surfaces				

M	easured by the	student:		Nai	
				1 (4)	
Ve	<b>erified</b> by the te	eacher:		Signature of t	the teacher
				-	ne teacher
No	ote: Graphics of	f field lines are en	nclosed with this	report.	
D.					
K	esume				
	est Questio				
		ion of the term: a			
		ion of the term: f	_	- C: -1.1	
			otential of electrical at a	given point of fie	ld mathematically
4.	connected?	neid stietigtii ai	iu potentiai at a	given point of the	id mamematicany
5.		ld lines and equir	otential surfaces	are mutually perpe	endicular.
		is called equipote		and manamily perpe	
		field is called po			
8.	How do electr	rode forms influe	nce the electric fi	eld?	
9.	What electric	field is called un	iform?		
Αı	nswers				

<b>Executed</b> by the student:	
Group	_
First name	
Last name	
<b>Approved</b> by the teacher:	
Date	Signature of the teacher
	$\mathcal{C}$

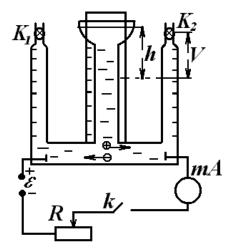
## Report on Laboratory Experiment No 2

## **Determination of the Charge of Hydrogen Ion**

		student:	
Group			
First name			
Last name			
is allowed to do the lab	oratory work.		
Date			Signature of the teacher
Purpose of Work			
•	f hydrogen ion and	compare it with	h the elementary charge.
Devices and Instru	ıments		
Hoffman's voltmeter, r		irect current, m	nilliammeter, switch,
connecting wires.			
Theoretical Conte	-		
The ions of hydrogen a sulphuric acid H <sub>2</sub> SO <sub>4</sub> .			rsis of aqueous solution of
Molecules of oxygen ap	ppear near the anode	e as a result of	the following process
Charge of ion can be de	etermined		
where $Q^+$		_ n <sup>+</sup>	
Calculation Form (with explanation of va	lues which are used	´ l -	
$q^+ = {2}$	$\frac{RTIt}{N_0 P_{H_2} V},$	$\begin{vmatrix} I - \\ t - \\ N_0 - \end{vmatrix}$	
V		T	

$$P_{H_2} = P_{am} + \rho g h - a P_t,$$

## **Experimental Assembly**



where		
$K_1, K_2 -$		
1 / 2		
R –		
k –	 	
mA –		
V		
ε		

### **Measurement Results**

$N_{\underline{0}}$	I	T	T	V	$P_{at}$	h	$P_h$	$P_t$	$P_{H2}$	$q^{^{+}}$
exp	[A]	[sec]	$[^{0}K]$	$[m^3]$	[kPa]	[m]	$P_h$ [kPa]	[kPa]	$P_{H2}$ [kPa]	[C]
1.										
2.										
3.										
4.										
				Avera	ge value	;				

<b>Measured</b> by the student:	
•	Name
Verified by the teacher:	
•	Signature of the teacher

## **Analysis**

Let's introduce experimental results into the formulas and calculate the charge of hydrogen on:

$$q_1^+ =$$

$$q_{2}^{+} =$$

$$\overline{q}^{\scriptscriptstyle +}$$
 =

Let's compare the calculated results with the value of elementary charge:

#### Resume

Absolute and ratio errors of the value of  $q^+$  for the experiment are:

$$\frac{\Delta \overline{q}}{\overline{q}} = \sqrt{\left(\frac{\Delta \overline{T}}{\overline{T}}\right)^2 + \left(\frac{\Delta \overline{I}}{\overline{I}}\right)^2 + \left(\frac{\Delta \overline{t}}{\overline{t}}\right)^2 + \left(\frac{\Delta \overline{P}_{H_2}}{\overline{P}_{H_2}}\right)^2 + \left(\frac{\Delta \overline{V}}{\overline{V}}\right)^2};$$

Basis of errors:

 $\Delta \overline{I} = 0.95 \cdot \Delta I =$ \_\_\_\_\_\_, where  $\Delta I = \frac{k}{100\%} \cdot I_{\text{max}} =$ \_\_\_\_\_\_

(k – grade of accuracy of device,  $I_{\text{max}}$  – limiting value of the device scale)

 $k = \underline{\hspace{1cm}}, \ I_{\max} = \underline{\hspace{1cm}}$ 

 $\Delta \bar{t}_{SM} = 0.95 \cdot \ell_t =$ \_\_\_\_\_\_\_, where  $\ell_t =$ \_\_\_\_\_\_\_, since value of t is measured\_\_\_\_\_\_ with accuracy\_\_\_\_\_\_

$$\Delta \overline{P}_{H_2} = \sqrt{(\Delta \overline{P}_{at})^2 + (\rho g \Delta \overline{h})^2 + (0.9 \Delta \overline{P}_T)^2}$$

 $\Delta \overline{P}_{at} = 0.95 \ell_{P_{at}} = ______,$  where  $\ell_{P_{at}} = ______,$  since value of  $P_{at}$  is measured\_\_\_\_\_\_ with accuracy\_\_\_\_\_\_

 $\Delta \overline{h} = 0.95\ell_h =$ \_\_\_\_\_\_, where  $\ell_h =$ \_\_\_\_\_\_, since value of h is measured\_\_\_\_\_ with accuracy\_\_\_\_\_

 $\Delta \overline{P}_T = 0.95\ell_P = \underline{\hspace{1cm}}$ , where  $\ell_P = \underline{\hspace{1cm}}$ ,

is defined in the following way:

at temperature  $(t+1)^0 C$  value of  $P_{t_1} =$  (from Table)

at temperature  $(t-1)^0 C$  value of  $P_{t_2} =$ 

$$\ell_{P_t} = \frac{P_{t_2} - P_{t1}}{2} =$$

 $\Delta \overline{V} = 0.95 \ell_V =$  \_\_\_\_\_\_, where  $\ell_V =$  \_\_\_\_\_\_, since value of V is measured \_\_\_\_\_ with accuracy \_\_\_\_\_

$$\frac{\Delta \overline{q}}{\overline{q}} =$$

$$\Delta \overline{q} =$$

<b>Final Result</b>
---------------------

$$\overline{q} + \Delta \overline{q} =$$
 with probability  $\alpha = 0.95$ .

#### **Test Questions**

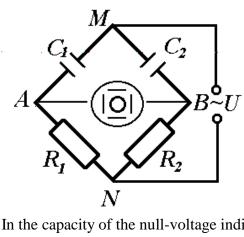
- 1. What is the value of elementary charge?
- 2. What is the effect of electrolytic dissociation?
- 3. Formulate the first Faraday law.
- 4. What charge type has the ion of hydrogen?
- 5. Where would be larger the volume of evolved gases, near cathode or anode? Why?
- 6. You mix the acid and water. The order of mixture: the first component is acid and water is the second. Why is such order used?
- 7. What influences the value of the electrolyte conductivity?

7. What influences the value of the electrolyte conductivity.				
Answers				
<b>Executed</b> by the student:				
Group	_			
First name				
Last name				
<b>Approved</b> by the teacher:				
Date	Signature of the teacher			

## **Report on Laboratory Experiment No 3**

## **Measurement of Capacitance and Permittivity by the Bridge Method**

The student:	
Group	
First name	
Last name	
is allowed to do	the laboratory work.
	<u></u>
Date	Signature of the teacher
Purpose of w	<b>ork:</b> capacitance measurement of two capacitors, capacitance of capacitors in series and parallel connection and permittivity of a substance.
Devices and i	<b>nstruments:</b> sound-frequency generator, oscillograph, two standard capacitors, two capacitors to be investigated, plane air capacitor, dielectric plates, variable resistor, trammel, ruler.
Theoretical	Contents. Summary
Permittivity of a	substance is



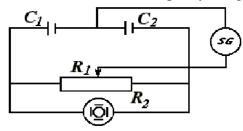
A bridge method of capacitance measuremen means	nt _

In the capacity of the null-voltage indicator	is being used.
For the reason	

will get  $C_1 = C_2 \frac{R_2}{R_1}$ ,

where  $C_2$  –

The electric scheme of a capacity bridge is shown:



To measure the permittivity of a substance $\varepsilon$	
	is used.

1 , ,	Table of symbols:
d <sub>o</sub>	1 –
$1$ $\mathbf{d}_2$	2 –
	$a_0 - \underline{\hspace{1cm}}$ $d_1 - \underline{\hspace{1cm}}$
2	$d_2$
The permittivity $\varepsilon$ is calculated by using the	capacitance bridge method
$\mathcal{E} =$	
where $C$	
$d_2$	
ε <sub>0</sub>	
S	
<i>d</i> <sub>1</sub>	

## **Results of the Measurements and Calculations of Capacitances**

Table 1.

				Table 1.
Investigated	$R_1$	$\overline{R_1}$	$\overline{R_2}$	Value of capacitance
capacitance	(Ohm)	(Ohm)	(Ohm)	(microfarad)
		, ,	, ,	
$C_{x1}$				
$C_{x1}$				
$C_{x2}$				
$C_{xP}$				
(in parallel)				
$C_{xS}$				
(in series)				
				1

## Measurement Results of the Permittivity of a Substance $\boldsymbol{\epsilon}$

Table 2.

Dielectric	$d_{0}$	$d_1$	$d_{2}$	$S_{2}$	C	$\varepsilon$
	(mm)	(mm)	(mm)	$(m^2)$	(picofarad)	

## **Calculations of the Capacitances**

$C_{x1} =$	$R_1 =$	$R_2 =$
$C_{x2} =$	$R_1 =$	$R_2 =$
$C_{xP} =$	$R_1 =$	$R_2 =$
$C_{xS} =$	$R_1 =$	$R_2 =$

Let's compare the results of measurements  $C_{xP}$  and  $C_{xS}$  with the calculations on the basis of the measured  $C_{x1}$  and  $C_{x2}$ .

Measurements

**Calculations** 

### **Calculations of the Permittivities of Substances**

 $\varepsilon_1 = \varepsilon_2 = \varepsilon_2 = \varepsilon_2$ 

**Comparison with the Table Results** 

Companison with the Table Results		
Table data	Data you've got	
Glass	$\mathcal{E} =$	
$\varepsilon = 5 - 7$		
Viniplast	$\varepsilon =$	
$\varepsilon = 2 - 2.5$		

## **Analysis of Measurements**

Let's calculate precision of measurements by using the estimation method of indirect measurements.

Formula construction for the uncertainty of measurement

•	
Evaluation of inaccuracy of measurements	

**Measurement's results:** 

$$C_{x1} = \overline{C}_x \pm \Delta C_{x1} =$$

with the probability  $\alpha = 0.95$ 

## **Ratio error of measurements:**

R	esume
Τe	est Questions
	Why is the capacitance of a capacitor larger than the capacitance of an electric conductor?
2.	Which parameters influence the permittivity of a substance?
	What preferences do you get using the bridge method for the measurements of capacitance and permittivity?
4.	How can you get minor geometric dimensions of the capacitors and a large capacitance simultaneously?
5	Why do we need to use capacitors in electric circuits?
	What purposes are series and parallel connection of the capacitors applied?
Aı	nswers
Ex	ecuted by the student:
	oup
	est name

Last name	
Approved by the teacher:	
 Date	Signature of the teacher

## **Report on Laboratory Experiment No 4**

## **Determination of High Value Resistance and Capacitance with the Help of Relaxed Oscillations**

The student:	
Group	
First name	
Last name	
is allowed to do the laboratory work.	
Date	Signature of the teacher
Purpose of Work	
capacitance measurement of two capaciton parallel connection. Do measurement for the	-
Theoretical Contents. Summary Relaxed oscillations are	
Relaxed oscillations are	
	The capacitor voltage-time dependence is shown in Fig. 1. The table of symbols:
b d f	$t_1$
	t <sub>2</sub>
	-
$U_{i_1}$ $t_1$ $t_2$	$U_{ig}$
	$U_q$
√ ⊥ <sub>Uq</sub> c e g	The principle of operation of the relaxed
a t	oscillation generator is consist in
Fig.	
According to the second Kirchhoft law $U_{\scriptscriptstyle 0}$ =	= $IR + U_c$ , where $U_0$

#### **Calculation Formulas**

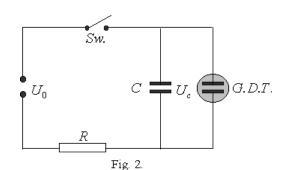
$$\frac{T_1}{T_2} = \frac{R_1}{R_2}$$
 ( $C = const$ ), where

$$\frac{T_1}{T_2} = \frac{C_1}{C_2}$$
 ( $R = const$ ), where

$T_1$ _	 	 <del></del>
$T_2$		

$$R_1$$

$$R_2$$



-- ectric Scheme

D.T	 	 

#### easurements results

Table 1.

Electric	The order of	The	t	T	$\overline{T}$	Unknown
circuit	experiment	numbers of	[sec]	[sec]	[sec]	quantity
parameters		flashes, n				1
$C_1 = R_1 = R_1$						
$R_1 =$						
1						
$R_1 =$						
$R_1 = C_{X1};$						$C_{X1} =$
$C_{X_1}$ ,						A 1
R -						$C_{X2} =$
$R_1 = C_{X2};$						$C_{X2}$ –
$C_{X2};$						
_						
$R_1 =$						
$R_1 = C_{XP};$						$C_{XP} =$
$R_1 =$						$C_{XS} =$
-				· · · · · · · · · · · · · · · · · · ·		

$C_{XS}$ ;			

Table 2.

						1 abie 2.
Electric	The order of	The	t	T	$\overline{T}$	Unknown
circuit	experiment	numbers of	[sec]	[sec]	[sec]	quantity
parameters		flashes, n				
$C_1 =$						
$C_1 = R_1 = R_1$						
1						
$C_1 =$						
$R_{X1}$ ;						$R_{X1} =$
X17						
$C_1 = R_{X2};$						$R_{X2} =$
$R_{v_2}$ :						
X 2 7						
$C_1 = R_{XP};$						
$R_{vp}$ ;						$R_{XP} =$
AF )						
$C_1 =$						
$C_1 = R_{XS};$						$R_{XS} =$
XS ?						

Measured by the student:	
•	Name
<b>Verified</b> by the teacher:	
•	Signature of the teacher

## **Analysis of Measurements**

Calculate ratio and absolute errors of only one series of measurement on teacher's option.

Formula for the error calculation:

$$\frac{\overline{\Delta C_X}}{\overline{C_X}} =$$

$$\frac{\overline{\Delta R_X}}{\overline{R_X}} =$$
**Note:** \text{ \text{Note: } \text{ \text{V}}

**Note:** Values of  $C_1$  and  $R_1$  are defined with the ratio equaling error 10%.

**Measurement's result:** 

$$C_X = \overline{C}_X \pm \overline{\Delta C_X} =$$

#### **Test Questions**

7. Can you explain the principle of operation of the relaxed oscillation's generator?

**8.** Get formula 
$$T = \ln \left( \frac{U_0 - U_{ig}}{U_0 - U_q} \right)$$
.

- **9.** What oscillations are called relaxed?
- 10. Formulate the second Kirchhoff's law.
- 11. Write down the differential equation of changing capacitor's charge.

<b>Answers</b>
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_	·	

<b>Executed</b>	by	the	student:
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Group	

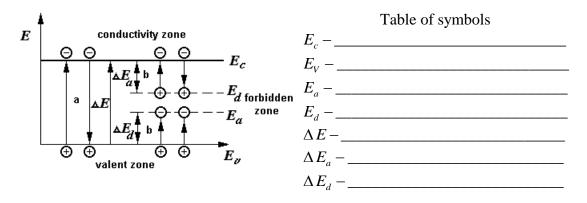
First name\_\_\_\_\_\_Last name\_\_\_\_\_\_

## **Approved** by the teacher: Date Signature of the teacher **Report on Laboratory Experiment No 5** Study of Semiconductor Resistance-Temperature **Dependence. Determination of Activation Energy** The student: Group First name\_\_\_\_\_ Last name\_\_\_\_ is allowed to do the laboratory work. Date Signature of the teacher Purpose of work Get the voltage-current characteristic and temperature-resistance dependence of a semiconductor. Define the activation energy for the given semiconductor. **Devices and Instruments** Semiconductor resistor (theristor), vessel with water, potentiometer, direct current sources, voltmeter, microammeter, switch and wires. **Theoretical contents. Summary** An activation energy of a semiconductor is \_\_\_\_\_ Two type of semiconductors exist in nature: *n-type* and *p-type*. These are schematically shown in Fig.1.

Ti~	1
riy.	

The role of impurity in semiconductor of <i>n-type</i> is			
The role of impurity in semiconductor of <i>n-type</i> is  The role of impurity in semiconductor of <i>p-type</i> is			
The role of impurity in semiconductor of <i>p-type</i> is			

On the empirical diagram:



The resistance of semiconductor decreases with the increasing of its temperature:

$$R =$$

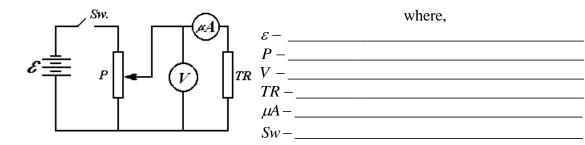
where, 
$$R_0$$
 —  $\Delta E$  —  $T$  —  $T$  —  $T$  dependence looks like \_\_\_\_\_\_

From this dependence it is possible to determine the value of \_\_\_\_\_

#### **Calculation Formulas**

$$R = \frac{U}{I}$$
 Where,  $U - \underline{\hspace{1cm}}$ 
 $I - \underline{\hspace{1cm}}$ 
 $\Delta E = k t g \varphi$   $t g \varphi - \underline{\hspace{1cm}}$ 
 $\Delta E - \underline{\hspace{1cm}}$ 

#### **Electric Scheme**



### **Results**

						T	able 1.
U							
(Volt)							
I							
(Ampere)							

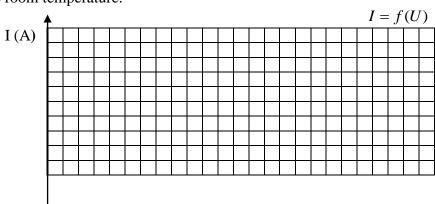
Table 2.

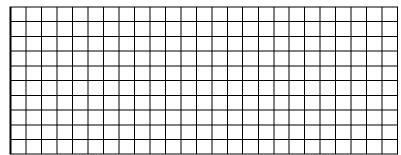
						Tuote 2.
$t^{o}C$	$T^{O}K$	$\frac{1}{T}, K^{-1}$	I (Ampere)	U (Volt)	R (Ohm)	ℓn R

Measured by the student:	
•	Name
Verified by the teacher:	
•	Signature of the teacher

## **Analysis**

1. Let's plot the graph of voltage-current characteristic using the data from Table 1 at the room temperature.

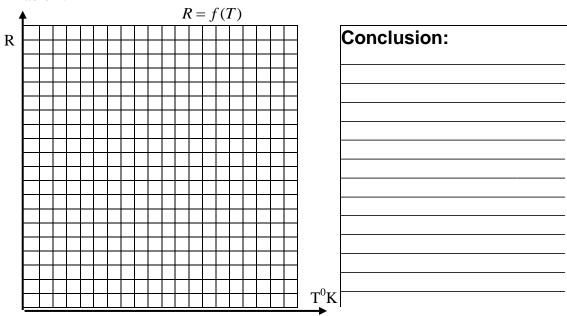




## Conclusions

\_\_\_\_\_

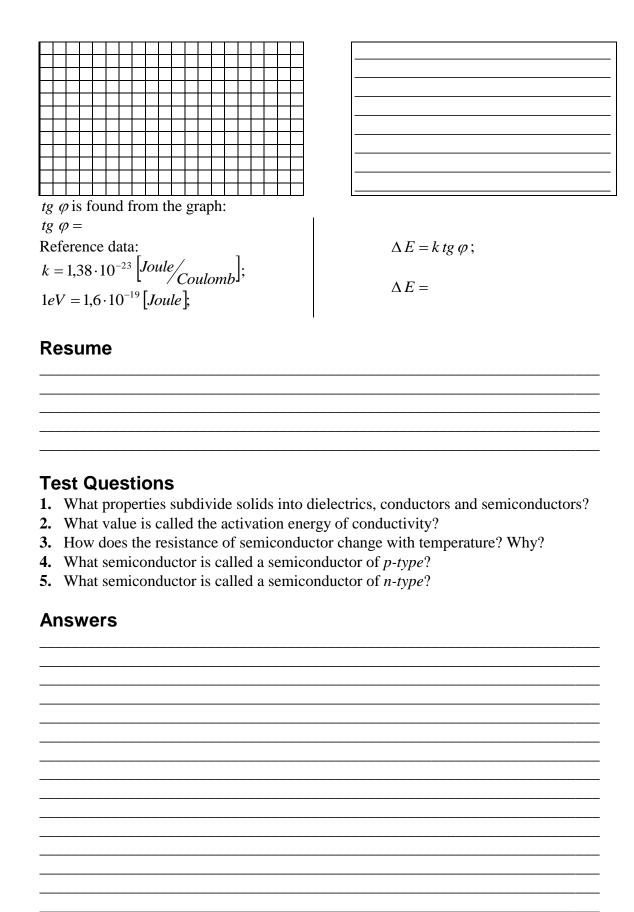
2. Let's plot the graph of temperature-resistance dependence using the data from Table 2.



3. Let's plot the graph  $\ell nR = f(\frac{1}{T})$  using the data from Table 2.

$\Gamma \ell$ 1	nk	_									
Н									_		
H									_		
Н											

Conclusion:



**Executed** by the student:

Group
First name
Last name
Approved by the teacher:
Date Signature of the teacher
Report on Laboratory Experiment No 6
Measurement of Magnetic Field of a Solenoid
The student:
Group
First name
Last name
is <b>allowed</b> to do the laboratory work.
Date Signature of the teacher
Purpose of Work  Determination of the magnetic induction and magnetic field strength inside a solenoid at the given points along it's axis. Determination of the magnetic induction dependence on the current, passing through the turns of the solenoid.  Devices and Instruments  Solenoid, millivoltmeter, measuring coil, rheostat and connecting wires.
Theoretical Contents. Summary
A solenoid is
The magnetic induction is
The magnetic field strength is

Method of mag	gnetic indu	ction meas	urement co	onsists in _			
Electric Cir	cuit of	the Exp	eriment	al Plant			
		L				e 1.1.	
Sw	•		$\mathfrak{M}$	A		of symbols	
<b></b> ✓ <b></b> [	$\Box$			L S -			
	R	D	~	Sw -			
•	<b>-(</b> A)-						
	_						
Calculation							
The magne $B =$	etic inducti	ion:					
where, $\mathcal{E}$ —				S _			
N							
4. Magnetic f	ield streng	th:					
H = where,							
$\mu_0$ –							
Results of	Measur	ements	and Ca	lculatio	ns		
	Mododi	011101110	and Ja				
I	1	1,5	2	2,5	3	3,5	Table 1
(Ampere)	-	_,-,-	<b>-</b>	_,_			-
$\mathcal{E}$ (Volt)							
B							
(tesla)							

H (Ampere/m)

 $\boldsymbol{x}$ 2 5 6 7 12 1 3 10 11 13 (Ampere) (*cm*)  $\overline{\mathcal{E}}$ (Volt)В (tesla) Н (Ampere / m) 19 14 15 16 17 18 20 21 22 23 24 25 26 (ampere) (*cm*)  $\mathcal{E}$ 

Table 2.

Measured by the student:	Name
Verified by the teacher:	Signature of the teacher

## **Analysis of Measurements**

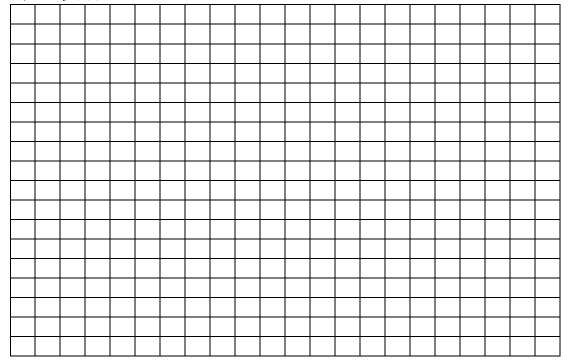
(volt)

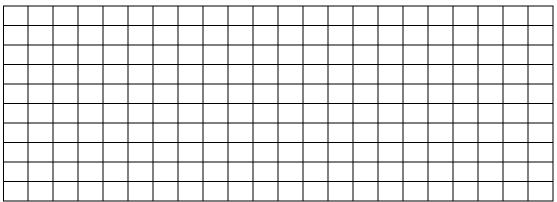
B
(tesla)

H
(ampere/m)

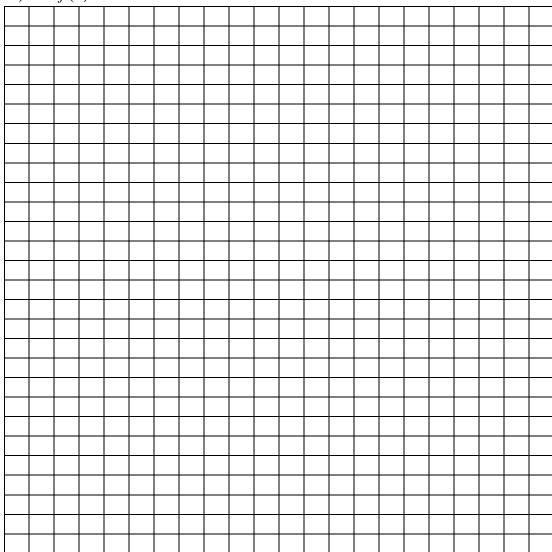
1. Plot the graphical dependence using the data from Table 1.

a) B = f(I);





- 2. Plot the graphical dependence using the data from Table 2.
  - b) B = f(x)



3. Calculate B using the data of measurements and Biot-Savart-Laplace law at three points on the solenoid's axis (the current strength is given by the teacher).

At the center point	B =
On the end side of solenoid	B =

Outside of solenoid $B =$
4. Compare experimental data of $B$ $\mu$ $H$ with the results of calculation at equastrengths of current.
5. Estimate the accuracy of measurements using an accuracy grade of devices.
$\Delta I = (accuracy.grade) \cdot 10^{-2} \cdot I_N =$
$\Delta \mathcal{E} = (accuracy.grade) \cdot 10^{-2} \cdot U_N =$
Resume
Test Overtions
<ul><li>Test Questions</li><li>1. What is the difference between the magnetic induction and magnetic field strength?</li><li>2. How does an induction depend on the magnetic strength?</li></ul>
3. What magnetic field is called uniform?
4. How does an induction in the central point change if the length of a solenoid i decreased?
5. What phenomenon is called an electromagnetic induction? How is it used in thi laboratory work?
Answers

<b>Executed</b> by the student:	
Group	_
First name	_
Last name	_
<b>Approved</b> by the teacher:	
Date	Signature of the teacher