## **TABLE OF PHYSICAL EFFECTS**

This table is a collection of physical effects. It helps in locating of various, sometimes rarely used, physical effects, which could help in solving the contradiction.

TEMPERATURE	POSITION and FLOW	POSITION CONTROL
<u>Measuring</u> <u>temperature</u> <u>Lowering</u> <u>temperature</u> <u>Raising temperature</u> <u>Stabilizing</u> temperature	Indication of position and location of object Measuring dimensions of objects <u>Control of aerosol flows</u> Changing the dimensions of objects <u>Controlling location of objects</u>	Control of movement Control of aerosol flows (dust, fog, smoke) Stabilization of position of object Action of forces. Control. Creation of high pressures Setting up interaction of mobile, (exchangeable), and immobile, (fixed), objects
SURFACE and VOLUME PROPERTIES	ENERGY TRANSFER and DESTRUCTION	ELECTROMAGN. / RADIATION / LIGHT
Changes in friction Checking of state and properties of surfaces Measuring surface properties Inspection of state and properties in volume Changing the volume properties of an object Creating a given structure. Stabilization of structure of an object	heat energy Transfer of energy <u>Action of forces</u>	Indications of electrical and magnetic fields Indications of radiation Generation of electromagnetic radiation Control of electromagnetic fields Controlling light, light modulation
MIXING, SEPARATING, CHEM. CONVER.	MISCELLANEOUS	
<u>Forming mixtures</u> <u>Separation of</u> <u>mixtures</u> Initiation and intensification of chemical changes	Geometrical effects	

DEMAND ACTIONS	PHYSICAL EFFECTS
Measuring temperature:	$\cdot$ Heat distribution and the change it causes in the object's internal
	frequency of vibration.
	$\cdot$ Thermo-electrical phenomena (mostly thermocouples).
	<ul> <li>Spectrum of radiation (including IR sensors).</li> </ul>
	• Changes in optical, electrical, magnetic properties of substances.
	$\cdot$ Move through the Curie point.
	• Hopkins effect.
	· Barkhausen effect.
Lowering temperature:	Phase transitions.
	· Joule-Thomson effect.
	· Rank Effect.

	• Magnetic calorie effect.
	Thermoelectric phenomena.
Raising temperature:	Electromagnetic induction.
	· Vortical currents.
	· Surface effect.
	· Dielectrical heating.
	Electronic heating.
	· Electrical charges.
	· Absorption of radiation by the substance.
	· Thermo-electrical phenomena.
Stabilizing temperature:	• Phase transitions (including the move through the Curie point).
Indication of position and	Introduction of marker substances;
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location of object:	• Transforming the external fields (luminescent traces) or creating
	their own fields,
	(ferromagnetic) and hence easily inspected.
	· Reflection and emission of light.
	· Photo effect.
	· Deformation.
	$\cdot$ X-ray and radioactive radiation.
	· Luminescence.
	· Change in electrical and magnetic fields.
	· Electrical discharges.
	· Doppler effect.
Controlling location of	• Action of magnetic field on object or on ferro-magnet linked to
objects:	the object.
objects.	,
	Action of electrical field on charged object.
	Transfer of pressure of liquids and gases.
	Mechanical oscillations.
	· Centrifugal forces.
	· Heat distribution.
	· Light pressure.
Control of movement:	· Capillary action.
	• Osmosis.
	· Toms effect.
	· Bernulli effect.
	· Wave movement.
	· Centrifugal forces.
	· Weissenberg effect.
Control of aerosol	• Use of electrical charges.
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flows (dust, fog, smoke):	Electrical and magnetic fields
Promotor en 1	· Light pressure.
Forming mixtures:	· Ultrasonics.
	Cavitation (including homogenizers).
	· Diffusion.
	· Electrical fields.
	$\cdot$ Magnetic field in conjunction with a ferromagnetic substance.
	· Electrophoresis.
	· Solubilization.
Separation of mixtures:	Electrical and magnetic separation.
F	• Changing apparent viscosity of the separator liquid under the
	effect of electrical and magnetic fields.
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	· Centrifugal forces.
	• Sorption.
	• Diffusion.
	• Osmosis.
1	• Flotation

	Pressure swing distillation
	Extractive distillation
Stabilization of position	Electrical and magnetic fields.
of object:	• Fixing in liquids which harden in magnetic and electrical fields.
	· Hygroscopic effect.
	· Reactive movement.
Action of forces. Control.	• Effect of a magnetic field via a ferromagnetic substance.
Creation of high	Phase transitions.
pressures:	• Heat distribution.
	· Centrifugal forces.
	$\cdot$ Changing the hydrostatic forces by changes in the apparent
	viscosity of magnetic
	or electrically conducting liquids in a magnetic field.
	· Use of explosives.
	· Electro-hydraulic effect.
	• Optical hydraulic effect.
	• Osmosis.
Changes in friction:	Johnson-Rabeck effect.
	Action of radiation.     Kragelely, phonomenon
	Kragelsky phenomenon.     Oscillation.
Destruction of object:	· Electrical discharges.
Destruction of object:	Electro-hydraulic effect.
	Resonance.
	· Ultrasonics.
	· Cavitation.
	Induced radiation.
Accumulation of	Elastic deformations.
mechanical and heat	• Hygroscopic effect.
energy:	Phase transitions.
Transfer of energy:	· Deformations.
	• Oscillations.
	· Alexandrov Effect.
	• Wave movement including electric shock waves.
	• Radiation.
	· Heat conductivity.
	• Convection.
	<ul> <li>Phenomenon of reflection of light, (light carriers).</li> <li>Induced radiation.</li> </ul>
	Electromagnetic induction.
	Superconductivity.
Setting up interaction of	• Use of electromagnetic fields (transition from "substance" to
mobile, (exchangeable),	"field").
and immobile, (fixed),	
objects:	
Measuring dimensions of	Measuring inherent frequency of oscillation.
objects:	• Applying and reading magnetic and electrical markers.
Changing the dimensions	· Heat distribution.
of objects:	• Deformation.
	Magnetic electro-striction.
	Piezoelectrical effect.
Checking of state and	• Electrical discharge.
properties of surfaces:	· Reflection of light.
	• Electronic emissions.
	• Moire effect.
	· Radiation.

Measuring surface	• Friction.	
properties:	· Absorption.	
properties.	Diffusion.	
	· Bauschinger effect.	
	· Electrical discharges.	
	Mechanical and acoustic oscillations.	
	· Ultraviolet radiation.	
Inspection of state and	• Introduction of "marker" substances transforming the external	
properties in volume:	fields (luminescent traces) or creating their own fields,	
F F	(ferromagnetic), dependent on the state and properties of the	
	substance under study.	
	$\cdot$ Changing the mean electrical resistance depending on the	
	structure and properties of the object.	
	Interaction with light.	
	• Electric and magnetic optical phenomena.	
	· Polarized light.	
	$\cdot$ X-ray and radioactive radiation.	
	$\cdot$ Electronic paramagnetic and nuclear magnetic resonance.	
	• Magnetic resilient effect.	
	$\cdot$ Move through the Curie point.	
	Hopkins and Barkhausen effects.	
	• Measuring the inherent frequency of oscillation of an object.	
	• Ultrasonics, the Moessbauer effect.	
	• The Hall effect.	
Changing the volume	$\cdot$ Changing the properties of liquids, (apparent viscosity, fluidity),	
properties of an object:	under the action	
	of electrical and magnetic fields.	
	· Heat action.	
	<ul> <li>Phase transitions.</li> <li>Ionization under the effect of an electrical field.</li> </ul>	
	· Ultraviolet, X-ray, radioactive radiation.	
	• Deformation.	
	• Diffusion.	
	Electrical and magnetic fields.	
	· Bauschlinger effect.	
	• Thermoelectrical, thermo-magnetic and magnetic-optical effects.	
	• Cavitation.	
	· Photochromatic effect.	
	· Internal photo effect.	
Creating a given	Interference waves.	
structure. Stabilization of	• Standing waves.	
structure of an object:	· Moire effect.	
	• Magnetic waves.	
	· Phase transitions.	
	• Mechanical and acoustic oscillations.	
	· Cavitation.	
Indications of electrical	· Osmosis.	
and magnetic fields:	• Electrical charging f bodies.	
	• Electrical discharges.	
	Piezo- and magneto-electrical effects.	
	· Electrets.	
	Electronic emissions.	
	Electro-optical phenomena.	
	Hopkins and Barkhausen effect.	
	· Hall effect.	

	• Nuclear magnetic resonance.
	· Gyromagnetic and magnetic optical phenomena.
Indications of radiation:	· Optical acoustic effect.
	· Heat distribution.
	· Photoeffect.
	· Luminescence.
	· Photoplastic effect.
Generation of	· Josephson effect.
electromagnetic	· Induced radiation.
radiation:	· Tunnel effect.
	· Luminescence.
	· Hann effect.
	· Cherenkov effect.
	Stokes and anti-Stokes effects.
Control of	· Screening.
electromagnetic fields:	$\cdot$ Changing state of environment, for instance, increasing or
	decreasing its electric conductivity.
	$\cdot$ Changing the form of the surface of bodies interacting with fields.
Controlling light, light	· Refraction and reflection of light.
modulation:	• Electrical and magnetic optical phenomena.
	· Photoelasticity.
	• The Kerr and Faraday effects.
	· The Hann effect.
	• The Franz-Kieldysh effect.
Initiation and	· Ultrasonics.
intensification of	· Cavitation.
chemical changes:	• Ultraviolet, X-ray, radioactive radiation.
	· Electrical discharges.
	· Shock waves.
	• Mycellarian catalysis.
	Quick reactions at high temperatures.
Geometrical Effects:	Mobius Strip
	· Rotating Hyperboloid