

Institute of Power Engineering

Department of Nuclear and Thermal Power Plants

# Turbines in NPP

Master's program **140800 “Nuclear Physics and Technologies”**,  
educational program specialization **“Operation of Nuclear Power Plant”**

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Assistant Professor, DNTPP

Lectures: 48 hours

Independent learning: 60 hours

Test

Recommended reading:

1. *Leyzerovich, Alexander. Wet-Steam Turbines for Nuclear Power Plants / A. Leyzerovich.* — Tulsa : PennWell, 2005. — 456 p.
2. *Singh, Murari. Blade Design and Analysis for Steam Turbines / M. P. Singh, G. Lucas.* — New York : McGraw-Hill, 2011. — 364 p.
3. *H. P. Bloch, M. P. Singh. Steam Turbines. Design, Applications and Rerating.* -
4. *Паровые и газовые турбины для электростанций. / А.Г. Костюк, В.В. Фролов, А.Е. Булкин, А.Д. Трухний; под ред. А.Г. Костюка.* — М.: Издательский дом МЭИ, 2008. 556 с.
5. *Турбины тепловых и атомных электрических станций. / А.Г. Костюк, В.В. Фролов, А.Е. Булкин, А.Д. Трухний; под ред. А.Г. Костюка и В.В. Фролова.* — М.: Издательство МЭИ, 2001. 488 с.

# Objectives of the course

**Nuclear power plant** is a station intended to produce **electrical energy** through conversion of **nuclear energy**

**Law of conservation of energy:**

energy does not disappear or reappear, it is converted

*How is electrical energy generated?*

Electrical energy is produced in a device called **Electric Generator**

**Generator** operating principle:

The First Law of Electromagnetic Induction:  
*EMF (electromotive force - electric current) is induced in the conductor which intersects magnetic field lines.*

The Second Law of Electromagnetic Induction:  
*An oppositely directed force impacts the conductor through which an electric current moves and which intersects magnetic field lines.*

$I$  – current ,  $a$ ;  $U$  – voltage,  $B$ ;

Power of the electric generator ( $W$ ):

$$N_{\mathcal{E}} = IU = 2\pi rnR = \omega M$$

$n$  – speed,  $s^{-1}$ ;

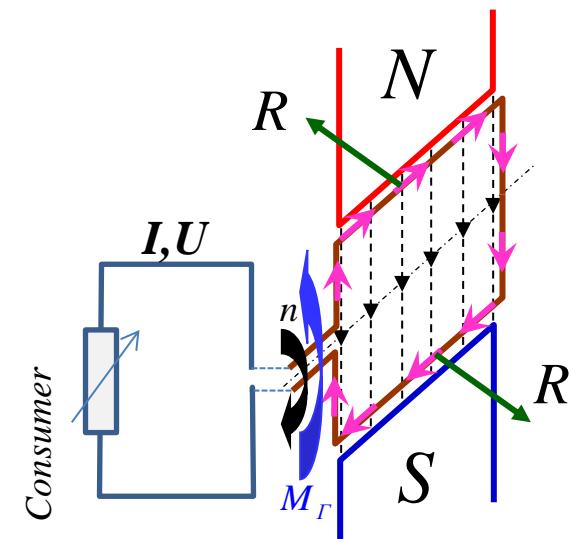
$r$  – frame radius,  $m$ ;

$2\pi rn$  – distance per unit time,  $m/s$ ;

$R$  – force,  $N$ ;

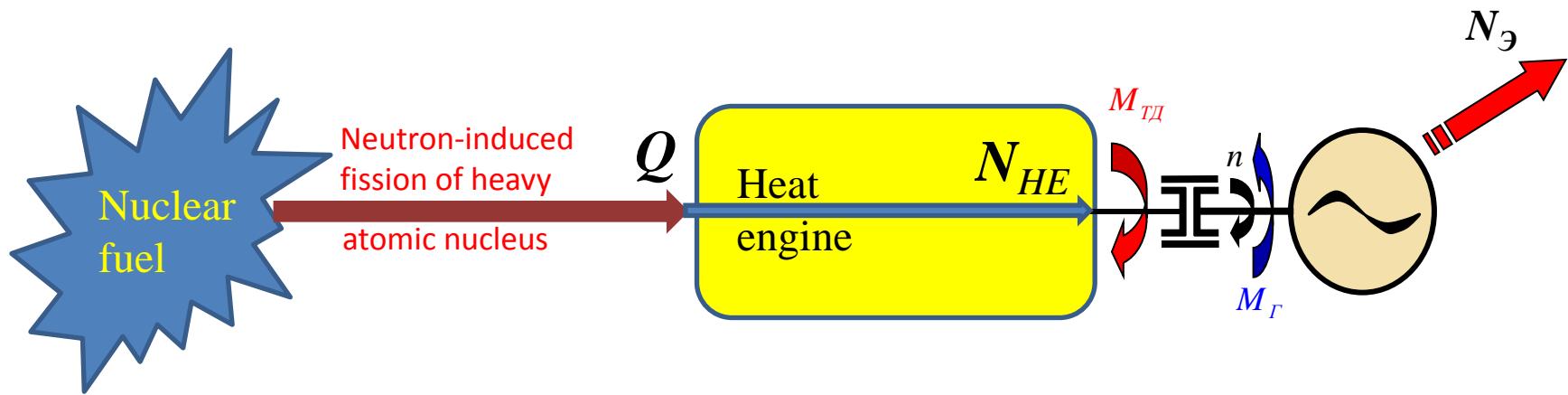
$rR=M$  – torque,  $Nm$ ;

$2\pi n=\dot{\omega}$  – angular velocity,  $s^{-1}$ .

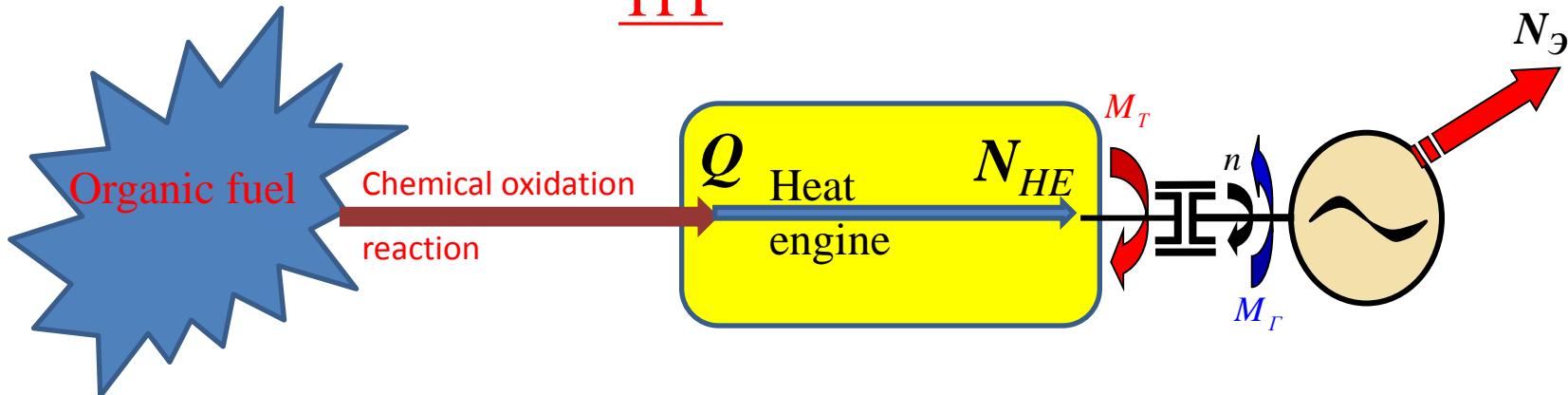


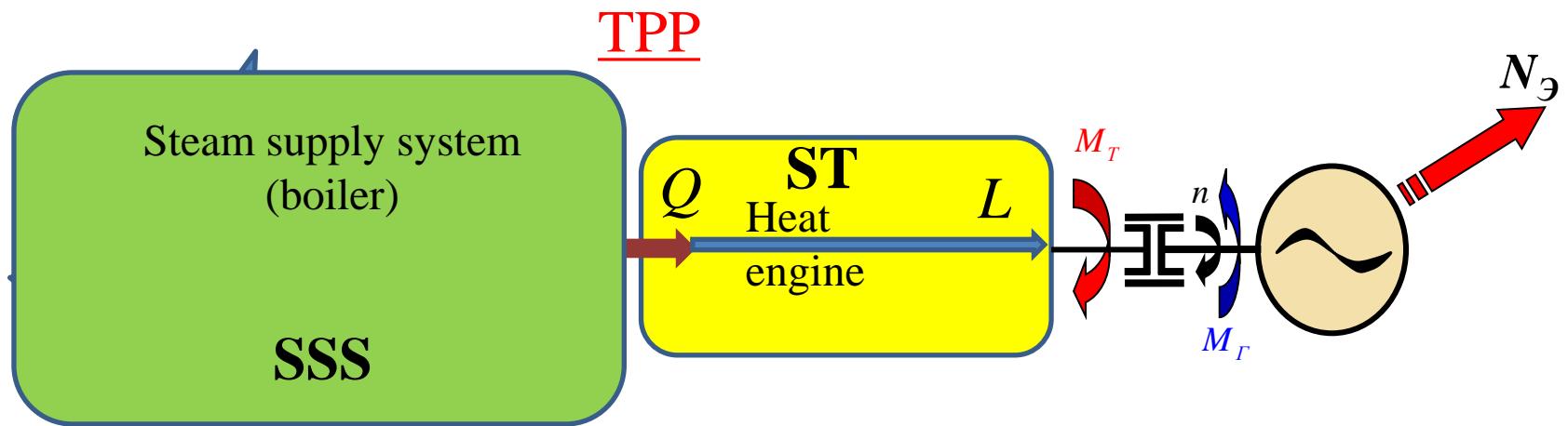
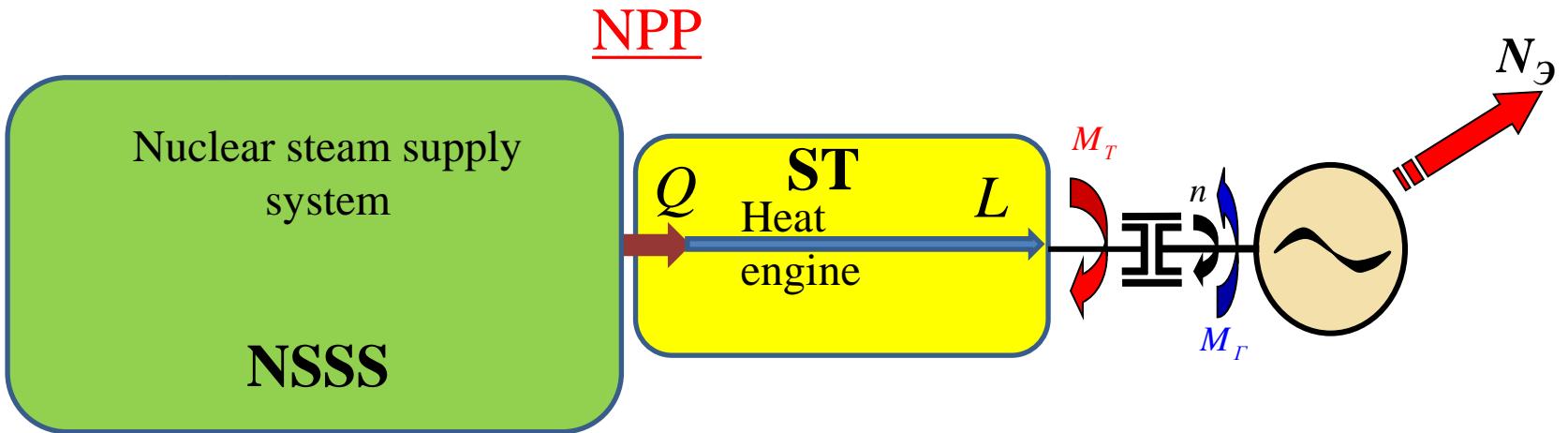
Conclusion: to produce electric current in the generator, the torque is to be **externally** applied to the generator rotor.

# Nuclear Power Plant (NPP)



TPP

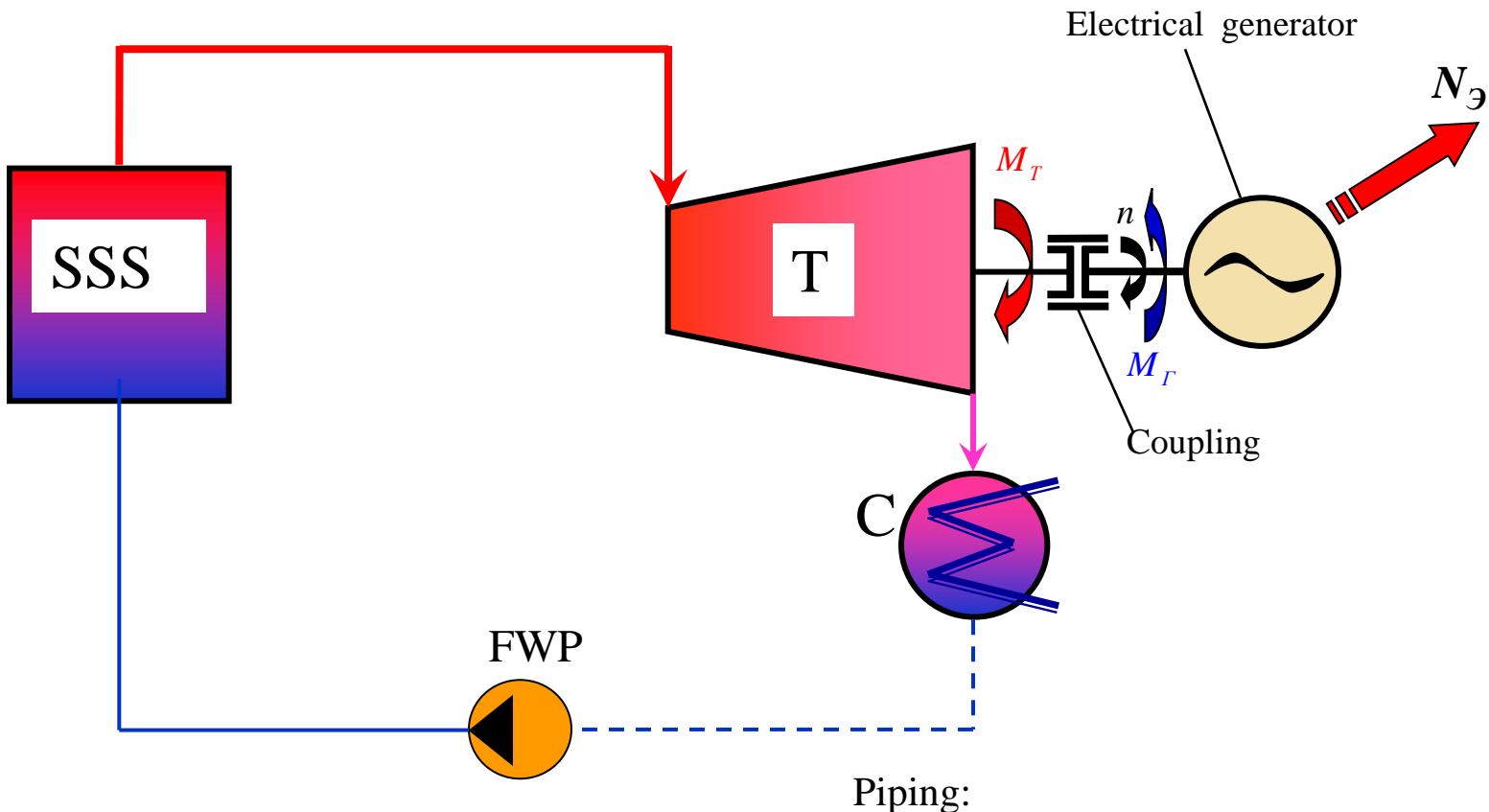




**ST** – steam turbine

**SSS** – steam supply system

# Diagram of a simple steam turbine



FWP *feedwater pump*

SSS *steam supply system*

T *turbine*

K *condenser*

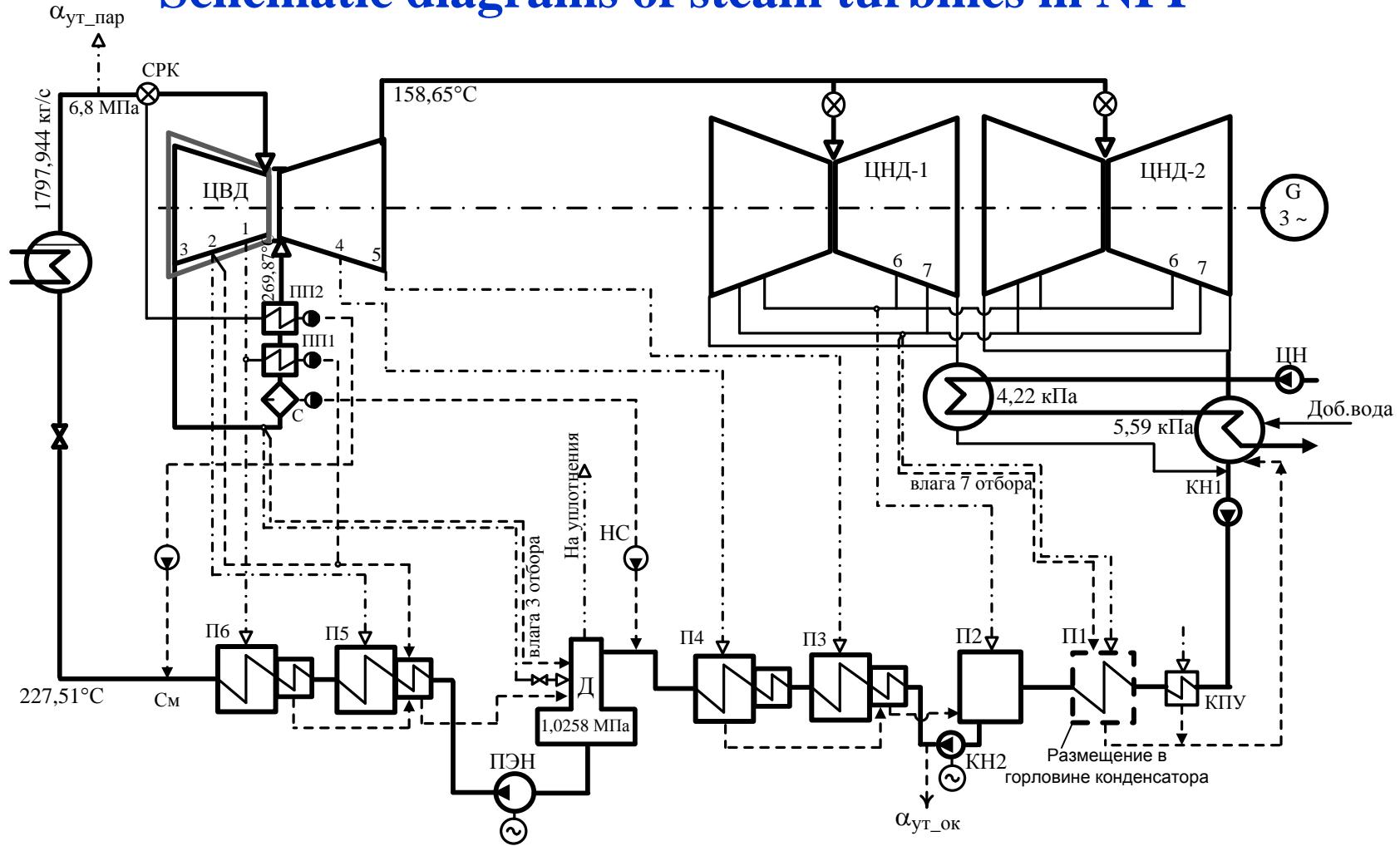
Piping:

- main steam
- Steam at the turbine outlet
- - - full-flow condensate
- feed water

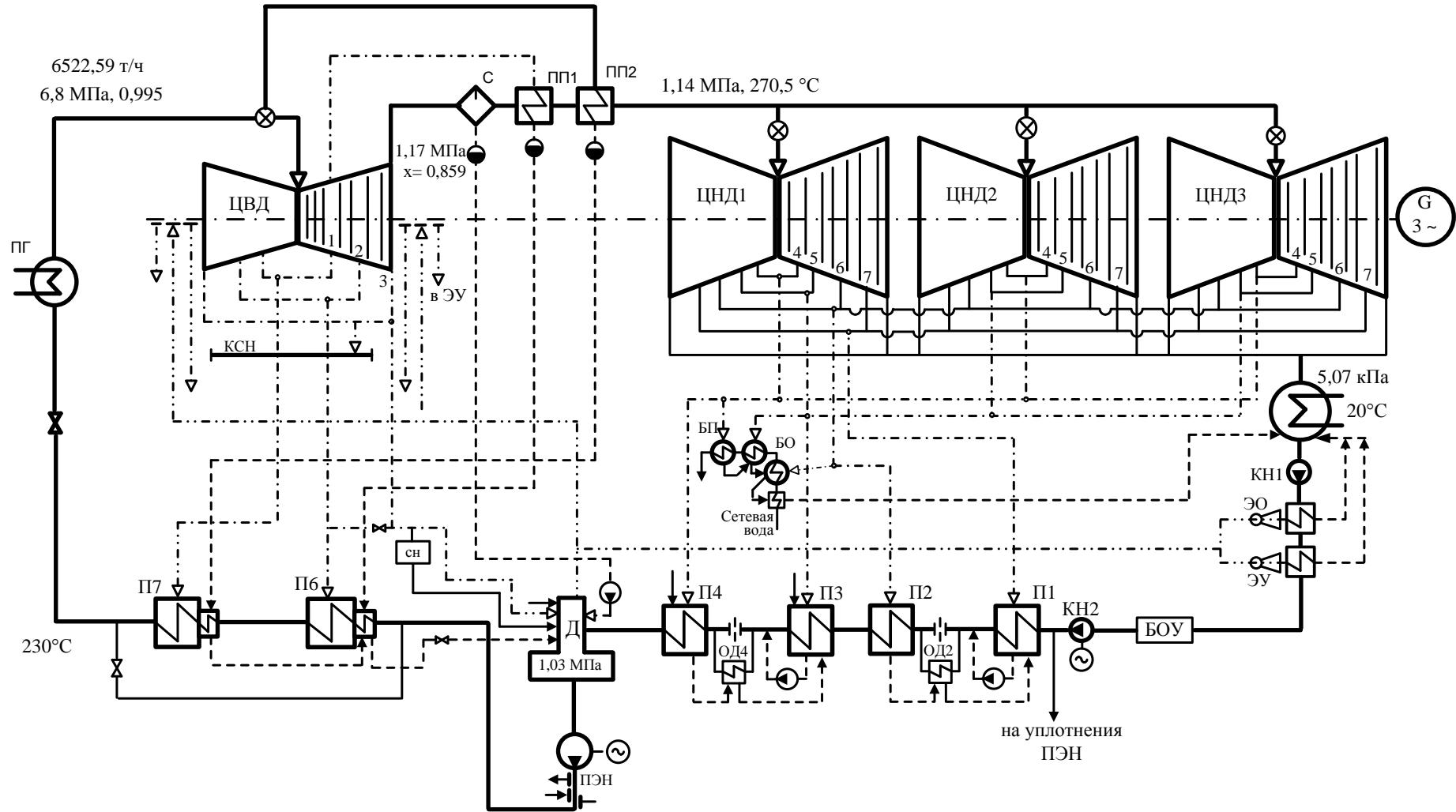
Basic requirements for steam turbines:

efficiency and reliability

## Schematic diagrams of steam turbines in NPP

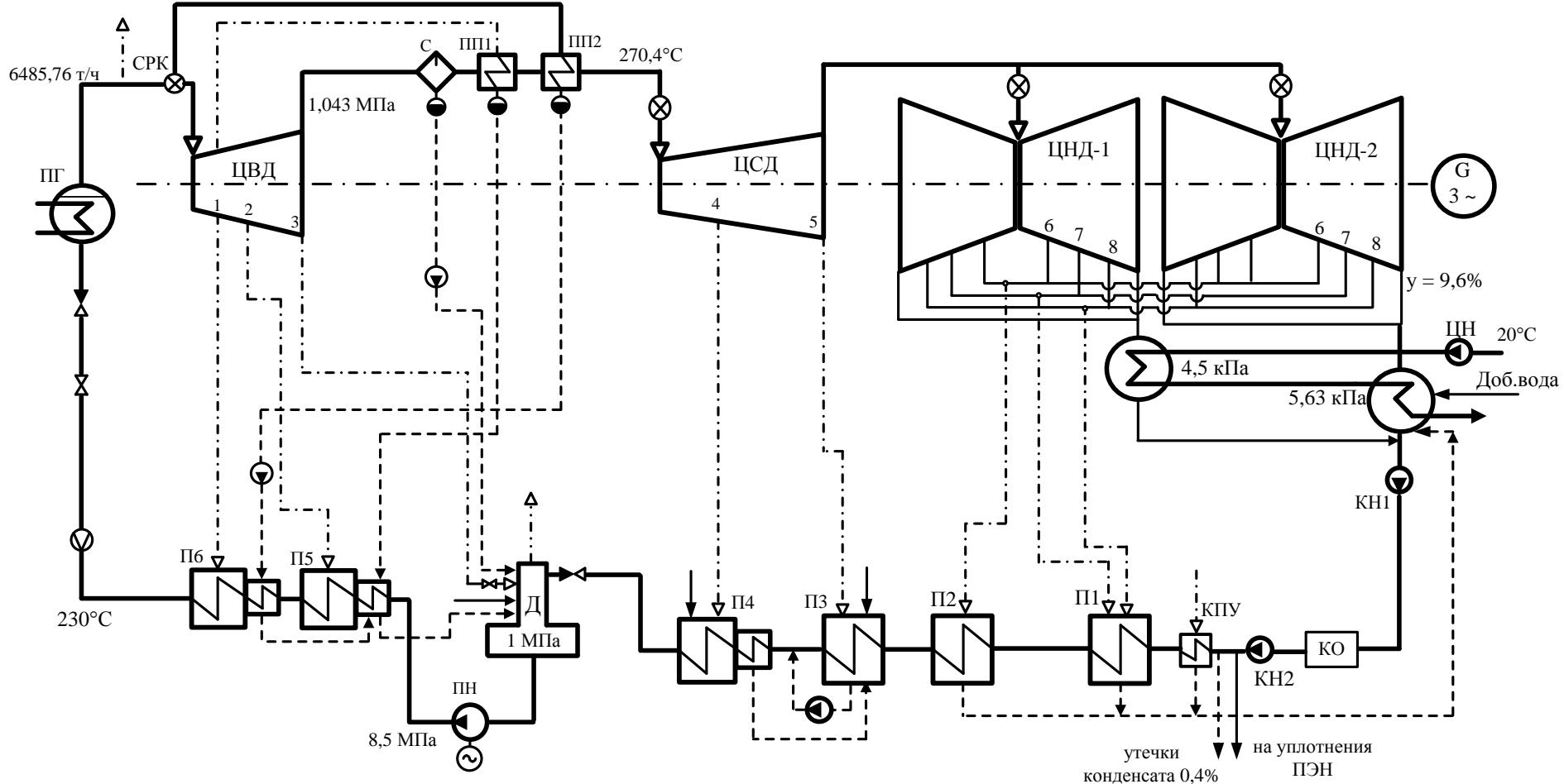


Cycle arrangement of the turbine K-1200-6.8/25 by JSC "Power Machines"



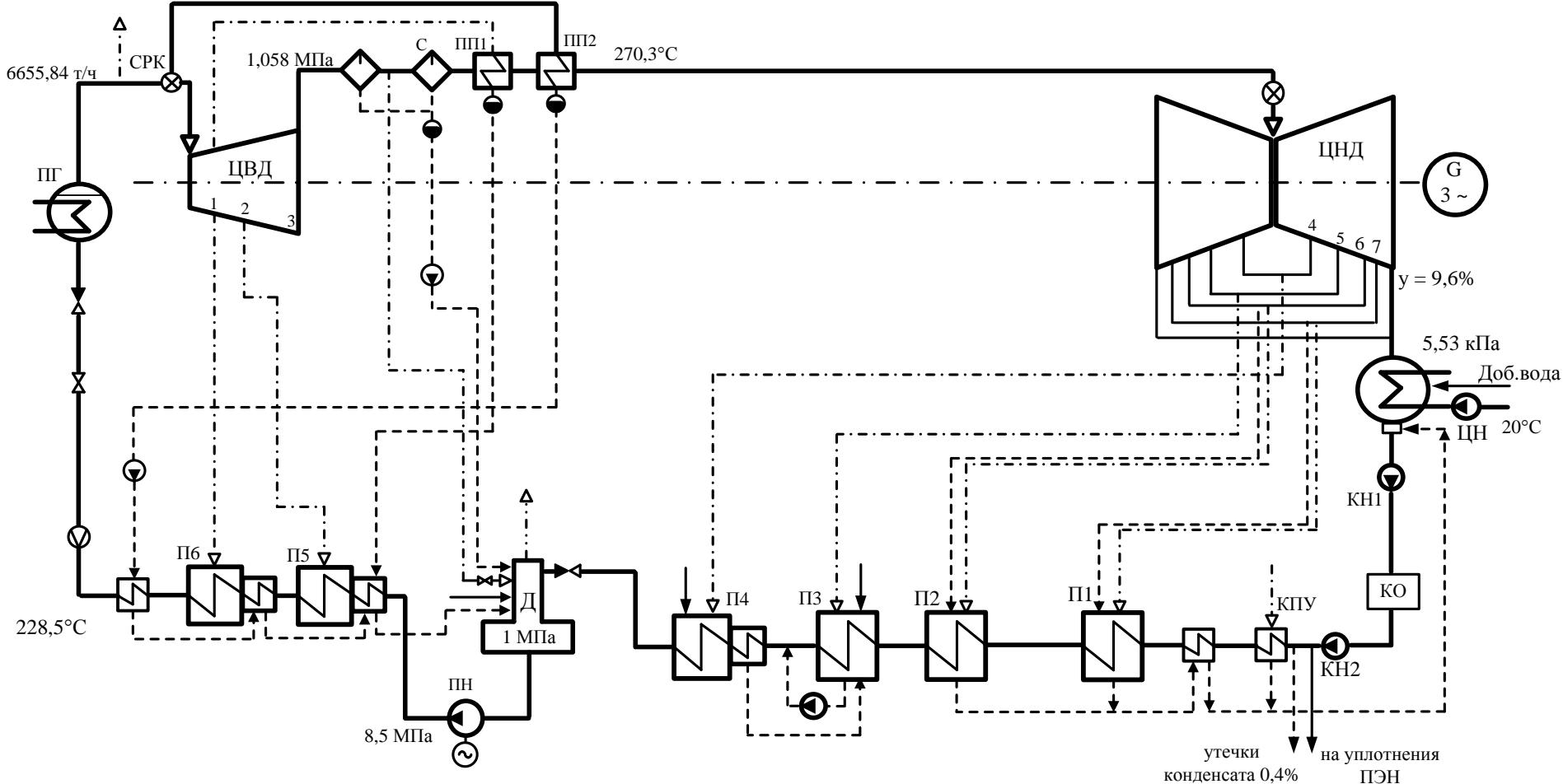
Cycle arrangement of the turbine K-1200-6.8/25 by JSC "Turboatom" (Kharkiv, Ukraine)

утечки пара 0,6%

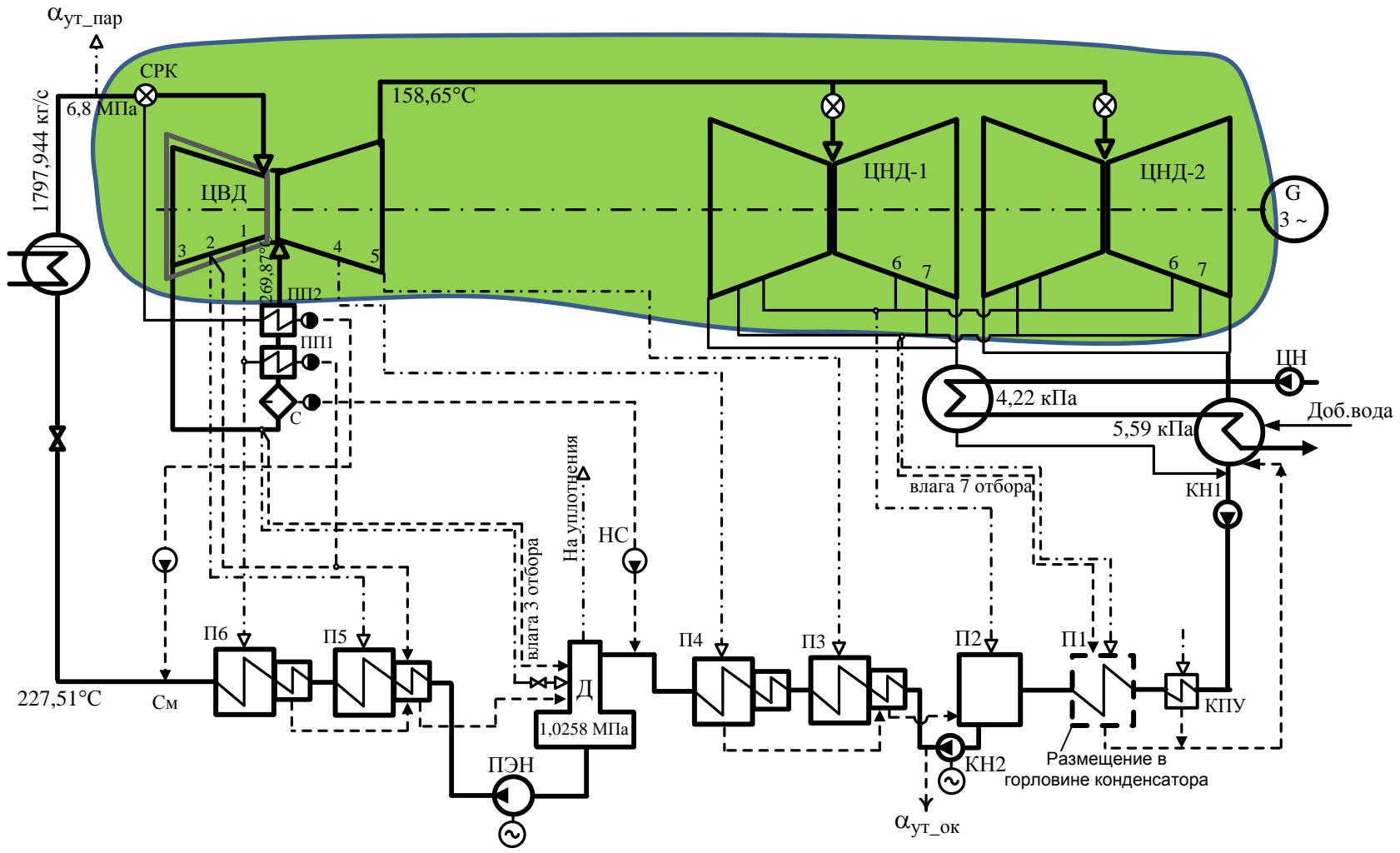


Cycle arrangement of the turbine by "ALSTOM"

утечки пара 0,6%

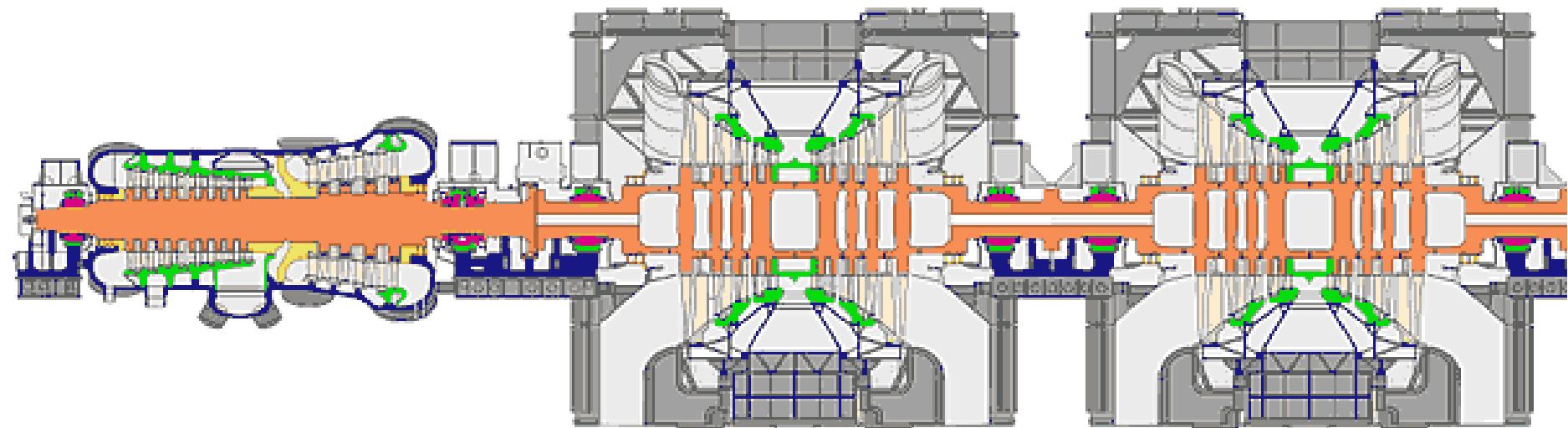


Information cycle arrangement of the turbine by "SIEMENS AG"



# Longitudinal cross section of the turbine K-1200-6.8/25, JSC "Power Machines"

$L_{last\ stage} = 1740$  mm



# State of matter

State of aggregation of matter: solid, liquid, gaseous, plasma.

Heating level: cold, warm, hot ...

Other characteristics...



How to quantify  
the state of  
matter?

The state of matter is unambiguously determined if we know  
**two *independent* thermodynamic properties...**

Thermodynamic properties:

- temperature – heating level of the body,  $t [^{\circ}C]$ ,  $T [K]$ 
  - internal energy  $u=f(t) [kJ/kg]$  is related to the temperature
- pressure is the exerted force of gas (or liquid) per unit area normal to this surface,  $p [Pa = N/m^2]$ ,  $kPa$ ,  $MPa$ ;  $bar$ ;  $ATA$ ;  $ATG$  ;

total, static, dynamic pressures are considered for the flow

- density is mass per unit volume,  $\rho [kg/m^3]$ 
  - specific volume is the volume of unit mass,  $v=1/\rho [m^3/kg]$
- enthalpy is stored energy per unit mass of the substance,  $h [kJ/kg]$
- entropy is the probability of the state of matter,  $s [kJ/kg \cdot K]$
- dryness fraction is the proportion of vapor in the wet steam volume .

$$ds = \frac{\delta q}{T}$$

Equations of state:

$$pV = RT \quad \text{Clapeyron equation}$$

$$h = \frac{k}{k-1} pV + \text{const} \quad \text{enthalpy equation,}$$

where  $k$  is specific heats ratio

$$k = \frac{c_p}{c_v}$$

✓ for superheated steam

$$k = 1,26 \dots 1,33$$

For estimations, it can be taken

$$k_{nn} = 1,3$$

✓ for dry saturated steam

$$k_{en} = 1,135$$

✓ for air

$$k_{go3d} = 1,4$$

The equations of state as formulas are used for **analytical description of processes**

***For calculations, we use:***

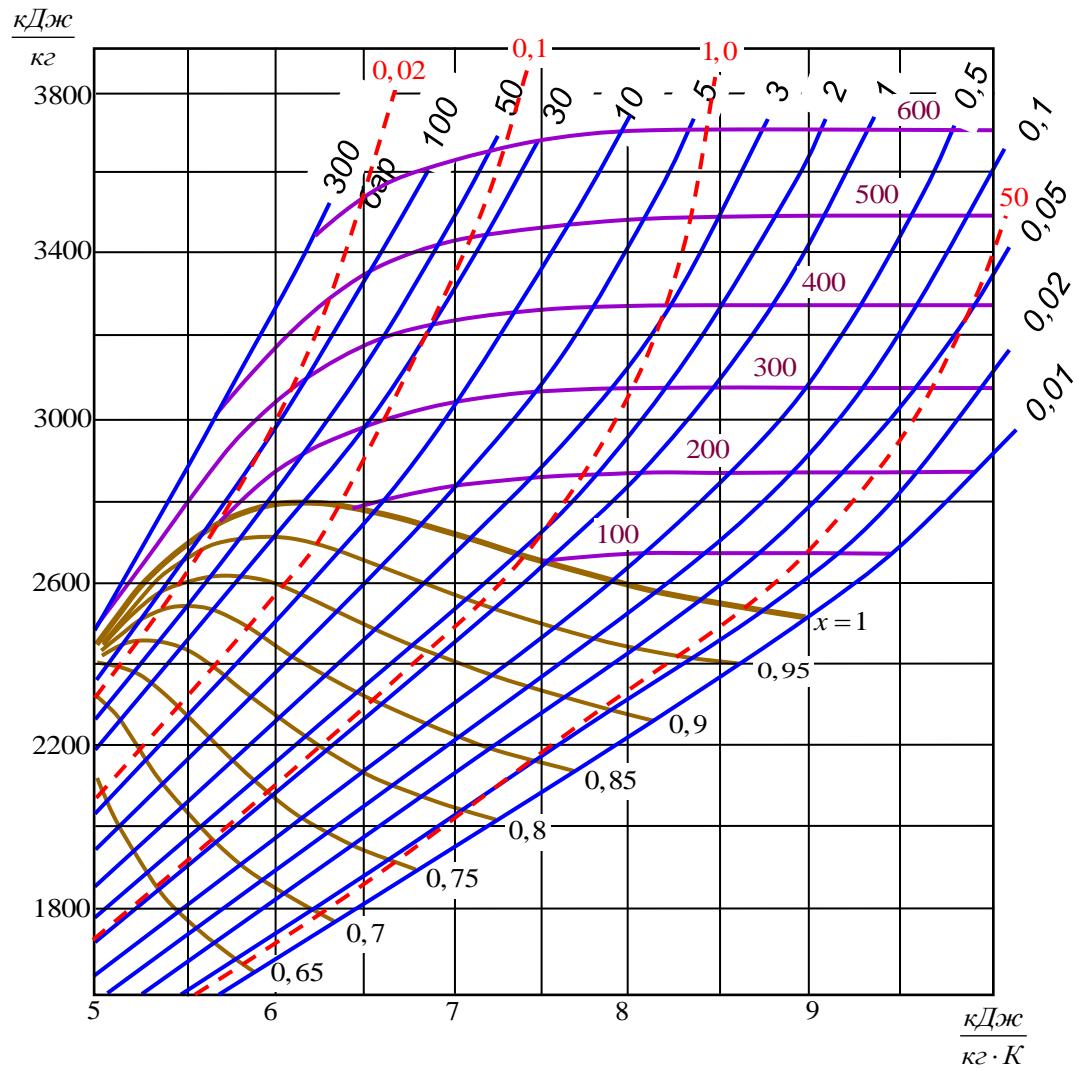
- **tables of thermodynamic properties; (discrete ratios of parameters)**
- **charts based on tables**

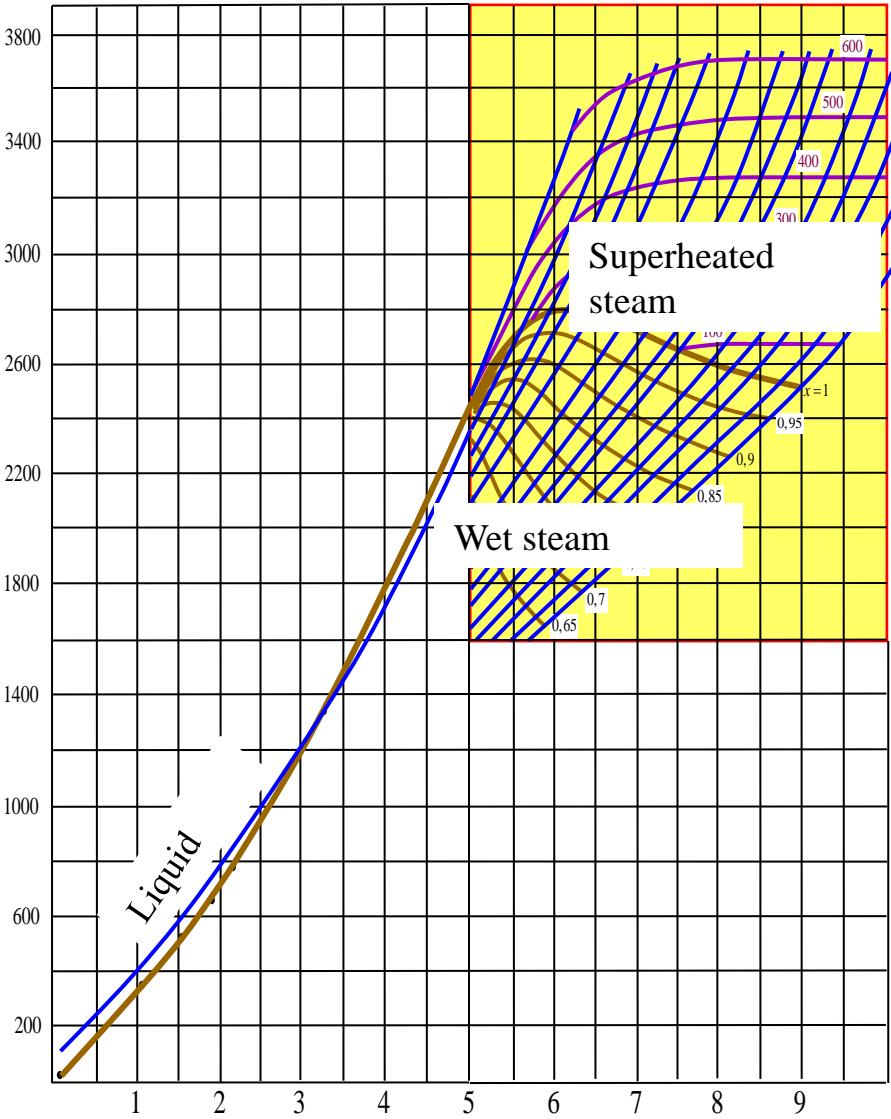
### A. hs Mollier chart

suggested in 1904 for calculation of steam turbines

The disadvantage is low accuracy of parameter determination

It is convenient in terms of process visualization

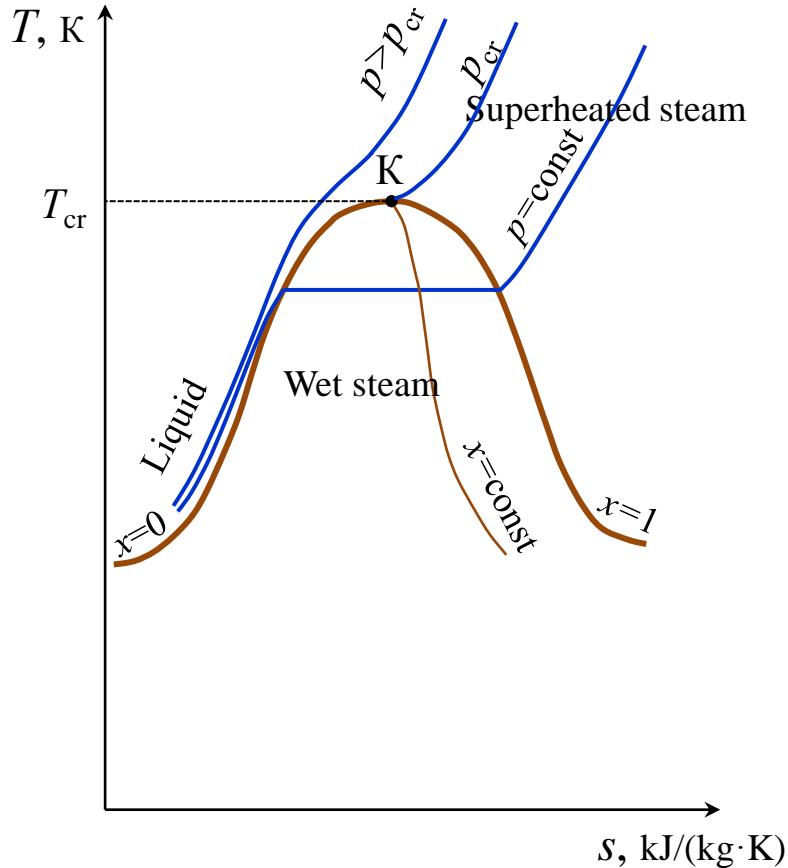




## B. Ts chart

p. K is critical point

$T_{cr}=375^{\circ}\text{C}$ ;  $p_{cr}=22.5 \text{ MPa}$



➤ electronic work sheets which are made based  
on approximation of thermodynamic properties

Change in the state occurs as a result of the process